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- Insecticidal substituted indazoles.
- © Compounds to the formula

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wherein Q is

$$(R_2)_n$$

A is an optionally substituted 1,2 or 3-atom bridge having 0 to 3 carbon atoms and 0 to 1 oxygen atom.

 NR_6 or $S(O)_q$ group;

B is H or an organic group;

J is H, alkyl or phenyl;

K is H or CH3;

R. and R2 are organic or inorganic substituents;

X is O or S;

Y is H or an organic substituent;

exhibit insecticidal activity and may be utilised for plant protection.

INSECTICIDAL SUBSTITUTED INDAZOLES

Cross-Reference to Related Application

This application is a continuation-in-part of U.S. Serial No. 036,480, filed on April 9, 1987.

Background of the Invention

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Field of the Invention

This invention pertains to substituted indazoles, agricultural compositions containing them and a method for using the indazoles to control insects.

State of the Art

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U.S. 4,070,365 discloses insecticidal compounds of the formula

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wherein

 R_2 and R_3 are independently alkyl, cycloalkyl, pyridyl or thienyl optionally substituted with halogen, alkyl or nitro, or a phenyl group optionally substituted with 1 or 2 substituents selected from halogen, alkyl, haloalkyl, cycloalkyl, alkylthio, alkoxy, mono or dialkylamino, nitro, phenyl optionally substituted with halogen, or cyano;

Y is halogen, NO₂, alkyl, haloalkyl, cycloalkyl, alkylthio, alkoxy, dialkylamino alkylsulfonyl, acyl, acylamino, cyano, or a phenyl optionally substituted with halogen; and X is O or S.

U.S. 4,663,341 discloses insecticidal compounds of the formula

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wherein

A is unsubstituted or substituted phenyl;

B is unsubstituted or substituted phenyl;

U is O. S or NR:

Y is alkyl, carbonyl or phenyl each of which may be optionally substituted; and Z is an organic radical other than hydrogen.

Summary of the Invention

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This invention pertains to compounds of Formula I, including all geometric and stereoisomers, agriculturally suitable salts thereof, agricultural compositions containing them and their use as insecticides:

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$$x \stackrel{c}{\overset{c}{\sim}}_{Y} \stackrel{(R_1)_m}{\longrightarrow}$$

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wherein Q is

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$$(R_2)$$

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A is a 1, 2 or 3-atom bridge having 0 to 3 carbon atoms and 0 to 1 oxygen atom, NR₅ group, or S(O)_q group, wherein each carbon individually can be substituted with 1 to 2 substituents selected from 1 to 2 halogen, C₁ to C₆ alkyl, C₂ to C₄ alkoxycarbonyl or phenyl optionally substituted with 1 to 3 substituents selected from W and one of the carbon atoms can be C(O) or C(S);

B is H, C₁ to C₆ alkyl, C₁ to C₆ haloalkyl, C₂ to C₆ alkenyl, C₂ to C₆ haloalkenyl, C₂ to C₆ alkynyl, OR₇, C₂ to C₆ alkoxyalkyl, C₂ to C₆ cyanoalkyl, C₃ to C₈ alkoxyarbonylalkyl, CO₂R₃, C(O)R₃, C(O)NR₃R₄, C(S)NR₃R₄, C(S)SR₃, phenyl, phenyl substituted by (R₅)_p, benzyl, or benzyl substituted with 1 to 3 substituents independently selected from W;

J is H, C₁ to C₄ alkyl or phenyl optionally substituted with W; K is H or CH₃;

W is halogen, CN, NO₂, C₁ to C₂ alkyl, C₁ to C₂ haloalkyl, C₁ to C₂ alkoxy, C₁ to C₂ haloalkylthio, C₁ to C₂ haloalkylthio, C₁ to C₂ haloalkylthio, C₁ to C₂ alkylsulfonyl or C₁ to C₂ haloalkylsulfonyl;

- 45 R₁, R₂ and R₅ are independently R₃, halogen, CN, N₃, SCN, NO₂, OR₃, SR₃, SOR₃, SO₂R₃, OC(O)R₃, OSO₂R₃, CO₂R₃, C(O)R₃, C(O)NR₃R₄, SO₂NR₃R₄, NR₃R₄, NR₄C(O)R₃, OC(O)NHR₃, NR₄C(O)NHR₃, NR₄SO₂R₃, or when m, n or p is 2, R₁, R₂ or R₅ can be taken together to form a 5 or 6 membered fused ring as -OCH₂O-, -OCH₂CH₂O-, or -CH₂CH₂O-, each of which can be substituted with 1 to 4 halogen atoms or 1 to 2 methyl groups;
- F₃ is H, C₁ to C₆ alkyl, C₁ to C₆ haloalkyl, C₂ to C₆ alkenyl, C₂ to C₆ haloalkenyl, C₂ to C₆ alkynyl, C₂ to C₆ alkoxyalkyl, C₂ to C₆ alkylthioalkyl, C₁ to C₅ nitroalkyl, C₂ to C₆ cyanoalkyl, C₃ to C₈ alkoxycarbonylalkyl, C₃ to C₆ cycloalkyl, C₃ to C₆ halocycloalkyl, phenyl, benzyl, or phenyl or benzyl substituted with 1 to 3 substituents independently selected from W;
- R₄ is H, C₁ to C₄ alkyl or R₃ and R₄ can be taken together as (CH₂)₄, (CH₂)₅ or (CH₂CH₂OCH₂CH₂);

 R₅ is H, C₁ to C₄ alkyl, C₁ to C₄ haloalkyl, C₂ to C₄ alkenyl, C₂ to C₄ haloalkenyl, phenyl optionally substituted with W:

 R_7 is H, C_1 to C_4 alkyl, C_2 to C_4 alkenyl, C_2 to C_4 alkylcarbonyl, C_2 to C_4 alkylcarbonyl, C_1 to C_4 alkylsulfonyl;

X is O or S;

n is 1 to 4:

m is 1 to 5:

p is 1 to 3;

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q is 0 to 2; and

Y is H. C. to C_6 alkyl, benzyl, C_2 to C_6 alkoxyalkyl, CHO, C_2 to C_5 alkylcarbonyl, C_2 to C_6 alkoxycarbonyl, C_2 to C_6 halo alkylcarbonyl. C. to C_6 alkylthio, C. to C_6 haloalkylthio, phenylthio, or phenylthic substituted with 1 to 3 substituents independently selected from W.

Preferred compounds (A) are those of Formula I wherein Q is

$$(R_2)_n$$
 $(R_3)_g$
 N
 N
 K
 $O-1$

$$(R_2)_n$$
 $(R_8)_q$
 $(R_8)_k$
 $(R_8)_k$

$$(R_2)_n \xrightarrow{(R_8)_g} J$$

$$N = N$$

$$0 = 3$$

$$(R_2)_n$$

$$N N K$$

$$Q-4$$

(R₂)_n Z_N

X is O or S;

X' is O or S;

t is 0, 1 or 2;

V is O, $S(O)_q$, or NR₆;

Z is O or NR₆.

 R_8 is H. halogen, C_1 - C_6 alkyl, C_2 - C_4 alkoxycarbonyl, phenyl or phenyl substituted by 1-3 substituents selected from W; and g is 0, 1 or 2.

Preferred compounds (B) are preferred compounds (A) wherein

Q-5

B is H, C₁ to C₆ alkyl, C₁ to C₅ haloalkyl, C₂ to C₆ alkenyl, C₂ to C₆ alkoxyalkyl, C₃ to C₈ alkoxycarbonylalkyl, CO₂R₃, C(O)R₃, C(O)NR₃R₄, C(S)NR₃R₄, C(S)R₃, C(S)SR₃, phenyl or phenyl substituted by $(R_5)_p$;

J is H or CH₃;

 R_3 is H, C₁ to C₂ alkyl, C₁ to C₂ haloalkyl, C₃ to C₂ alkenyl, C₃ to C₂ haloalkenyl, propargyl, phenyl, benzyl, or phenyl or benzyl substituted with one of F, Cl. Br. CF₃ or NO₂:

Rs is H, C+ to C4 alkyl, allyl or propargyl;

5 n is 1 or 2;

p is 1 or 2;

m is 1 to 3; and

Y is H, C+ to C₄ alkyl, SCH₃, SCCl₃, SO₂CH₃, SC₆H₅, 2-NO₂-C₅H₄S. C(O₁CH₃, C(O₁CF₃: CO₂CH₃ or CO₂CH₂CH₃.

Preferred compounds (C) are preferred compounds (B) wherein

B is H, C₁ to C₄ alkyl, C₃-C₄ alkoxycarbonylalkyl, CO₂R₃, C(O)R₃, phenyl or phenyl substituted by $(R_5)_p$, J is H;

K is H;

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R₁ is H, halogen, CN, SCN, NO₂, OR₃, SR₃, SO₂R₃, CO₂R₃, C(O)R₃ or R₃ with one R₁ substituent in the 4position, or when m is 2 then R₁ can be taken together as -CH₂C(CH₃)₂O-. OCH₂CH₂O-, OCF₂CF₂O-or -CF₂CF₂O-to form a 5 or 6 membered fused ring;

 R_2 and R_5 are H, R_3 , halogen, CN, SCN, NO2, OR3, SR3, SO2R3, OC(0)R3, OSO2R3, CO2R3, C(0)R3, C-(0)NR3R4, SO2NR3R4 or NR3R4;

R₃ is C₁-C₄ alkyl, C₁-C₂ haloalkyl, C₃-C₄ alkenyl or propargyl;

20 R4 and R8 are H or CH3;

X is O;

X' is O:

m is 1 or 2;

t is 1; and

25 q is O.

Preferred compounds (D) are preferred compounds (C) wherein

R₁ is Cl, F, Br, CF₃, CN, OCF₃, OCF₂H or SCF₂H;

 R_2 is H. Cl, F. Br, CN, CF_3 , CH_3 , OCF_2H , OCF_3 , SCH_3 , SCF_2H , SO_2CH_3 or NO_2 :

R₆ is H or CH₃;

30 B is H. CH₃, CH₂CH₃, CO₂CH₃, CO₂CH₂CH₃, phenyl or phenyl substituted by halogen. CN, CF₃ or NO₂, and:

Y is H.

Preferred compounds (E) are preferred compounds (D) wherein

Q is Q-1.

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Preferred compounds (F) are preferred compounds (D) wherein

Q is Q-2;

Specifically preferred compounds are preferred compounds (E):

Methyl 3,3a,4,5-tetrahydro-2-[[4-(trifluoromethyl)phenylamino]carbonyl]-2H-benz[g]indazole-3a-carboxylate, (G)

 Methyl 7-chloro-3,3a,4,5-tetrahydro-2-[[4-trifluoromethyl)phenylamino]carbonyl]-2H-benz[g]indazole-3a-carboxylate, (H)

7-Chloro-3,3a,4,5-tetrahydro-N-[4-(trifluoromethyl)phenyl]-2H-benz[g]indazole-2-carboxamide, (I)

Methyl 6-fluoro-3,3a,4,5-tetrahydro-2-[[4-(trifluoromethyl)phenylamino]carbonyl]-2H-benz[g]indazole-3a-carboxylate, (J).

The compounds of this invention, in addition to being useful as foliar and/or soil insecticides, can be combined with other pesticides for agricultural uses that will readily occur to one skilled in the art. Reference to compounds of this invention or to compounds of Formula I includes geometric and stereoisomers thereof as well as agriculturally suitable salts which, for brevity only, are not referred to specifically hereafter.

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Details of the Invention

For descriptive purposes, the following numbering system will be used to describe the values of Q in both the discussion and tables that follow.

Compounds of Formula II can be prepared by the reaction of aryl isocyanates of Formula II and substituted indazoles of Formula III as shown in Scheme I. Typical reactions involve the combination of equimolar amounts of II and III in conventional organic solvents including ether, tetrahydrofuran (THF), methylene chloride, chloroform and benzene. The reaction can be run at temperatures ranging from about -20°C to 100°C with temperatures in the range of about -10° to 30°C generally preferred.

SCHEME 1

$$(R_1)_{\underline{m}} \longrightarrow NCO + (R_2)_{\underline{n}} \longrightarrow R$$

$$NCO + (R_2)_{\underline{n}} \longrightarrow R$$

$$R_1 \longrightarrow R$$

$$R_1 \longrightarrow R$$

$$R_2 \longrightarrow R$$

II III

Substituted indazoles of Formula III, where B is equal to H, can be prepared by the reaction of hydrazine with an α , β -unsaturated ketone of Formula IV (or their precursors) by procedures well documented in the chemical literature (Scheme 2). For a review of pyrazoline synthesis, see El-Rayyes and Al-Awadi. Synthesis, 1028 (1985). For literature describing the synthesis of 3,4-and 3,5-disubstituted pyrazolines, which can be applied to the synthesis of compounds of Formula III, where B is H, see U.S. Patent 3,991.073 and U.S. Patent 4.070,365.

SCHEME 2

$$(R_2) \xrightarrow{n} K \xrightarrow{NH_2NH_2} III$$

$$(B-H)$$

Preparation of compounds of Formula I, where B is other than H, OR₇, phenyl or substituted phenyl can be achieved by metallation of the 3a-position of Formula I (where B is H) followed by reaction with a suitable electrophile as depicted in Scheme 3. Procedures for the related transformation of 4-substituted pyrazolines to 4.4-disubstituted pyrazolines are reported in European Patent Application 153127 and can be applied to substituted indazoles of Formula I. Metallation can be accomplished by deprotonation with a strong base, in a suitable solvent and at temperatures ranging from -78°C to 100°C. Useful base for this reaction includes lithium dialkylamides such as lithium diisopropylamide and lithium tetramethylpiperide, alkyl lithium reagents such as n-butyllithium and metal hydrides such as sodium hydride and potassium hydride. Deprotonation of compounds of Formula I, where B is equal to hydrogen, may require two equivalents of base when Y is equal to hydrogen. The reaction can be conducted in many conventional organic solvents and in certain instances a cosolvent may be useful. Suitable solvents include diethylether, tetrahydrofuran, tetrahydropyran, dimethylformamide, hexamethylphosphoramide, benzene and the like. Suitable electrophilic reagents for reaction with the metallated Formula I compounds include alkyl and substituted alkyl halides, alkyl chloroformates, acyl halides, isocyanates, dialkyl carbamoylhalides and

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related electrophiles which will be known to those skilled in the art.

SCHEME 3

where E+ is an electrophilic reagent.

An alternative procedure for introduction of the 3a-substituent, which in certain instances may be preferred over that of Scheme 3 due to higher yields and/or greater ease of synthesis, proceeds via the intermediacy of Compound V, wherein the 2-nitrogen has been derivatized with a suitable protecting group. Deprotonation with a strong base such as lithium disopropylamide, typically in stoichiometric quantities, followed by reaction with any of the previously described electrophiles provides compounds of Formula V, where B is other than hydrogen, OR₇, phenyl or substituted phenyl. Removal of the nitrogen protecting group provides the required Formula III intermediate. Nitrogen-protecting groups are well documented in the chemical literature, as are procedures for their preparation and cleavage. Examples include acetyl, trifluoroacetyl, benzoyl, substituted benzoyl, alkoxycarbonyl, benzyl and substituted benzyl. For a review

see Greene "Protective Groups in Organic Synthesis" (New York: John Wiley and Sons, 1981) pp. 218 to 287.

SCHEME 4

(B is other than H. OR₇. Ph or substituted Ph)

where E⁺ is an electrophile and PG is representative of a suitable nitrogen-protecting group.

Formula III compounds where B is equal to hydroxy are prepared by the reaction of hydrazine with a suitably substituted epoxide of Formula VI (Scheme 5, Reaction 1). The epoxides are available from the enones of Formula IV by procedures known in the art, the most widely employed method being epoxidation via the use of alkaline hydrogen peroxide (see, e.g., Wasson and House, Org. Syn., Coll. Vol. 4, 552 (1963)).

Compounds of Formula III where B is equal to OR₇ and R₇ is other than hydrogen, may be prepared from the Formula III compounds where B is equal to hydroxy via reaction with electrophilic reagents such as alkyl halides, acyl halides, alkyl chloroformates, chlorosulfonates, isocyanates and dialkyl carbamoylhalides (Scheme 5, Reaction 2). Those skilled in the art will recognize that conditions necessary for this

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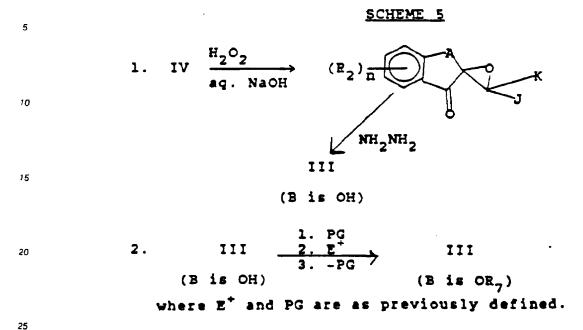
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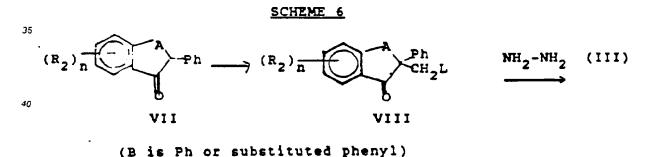
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transformation including solvent, catalyst, temperature, etc., will vary with the specific electrophile chosen and that protection of the 2-nitrogen may be necessary.



Compounds of Formula III where B is phenyl or substituted phenyl may be prepared by reaction of a substituted ketone of Formula VIII with hydrazine. The Formula VIII compounds are in turn prepared from compounds of Formula VII in a two-step process involving first an aldol or Mannich type reaction on the Formula VII compound followed by conversion of the hydroxyl or dialkylamino group to a leaving group L. examples of which include conversion to the chloride, bromide, tosylate or mesylate by procedures well known in the art.



Compounds of Formula IV (or their precursors) are readily prepared from compounds of Formula IX via an aldol condensation followed by dehydration of the aldol, or when J and K are hydrogen via a Mannich reaction followed by elimination of the dialkylamino group. For a general review of these reactions, see March, "Advanced Organic Chemistry", 3rd ed., (New York: John Wiley and Sons, 1985) pp. 829 to 834, and House, "Modern Synthetic Reactions", 2nd ed., (Menlo Park, CA: N.A. Benjamine, 1972) pp. 655 to 660.

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SCHEME 7

1.
$$(R_2)$$
 1. aldol condensation IV

2. IX Mannich (R₂)
$$\stackrel{R}{n}$$
 $\stackrel{NR}{}_{2}$ $\stackrel{IV}{\longrightarrow}$ $\stackrel{NR}{\longrightarrow}$ $\stackrel{IV}{\longrightarrow}$ $\stackrel{NR}{\longrightarrow}$ $\stackrel{IV}{\longrightarrow}$

Those skilled in the art will recognize the intermediates of Formula IX, which include compounds such as indanones, tetralones, chromanones, thiochromanones, and the like are well documented in the chemical literature as are procedures for their preparation. Selected examples of Formula IX are available commercially as well. The following Examples further illustrate the invention.

Example 1

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3,3a,4,5-Tetrahydro-N-[4-(trifluoromethyl)phenyl]-2H-benz[g]indazole-2-carboxamide

A mixture of 10.0 g of α -tetralone, 2.0 g of paraformaldehyde, 6.5 g of dimethylamine hydrochloride and 1.75 ml of conc. HCl in 20 ml of ethanol was heated at reflux for 24 h. cooled to room temperature and then partitioned between ether and water. The aqueous extracts were made basic with 1N aqueous NaOH and extracted with ether. The ether extracts were dried over magnesium sulfate and concentrated to 12.3 g of a yellow oil. The residual oil was dissolved in 40 ml of n-propyl alcohol, combined with 6.7 ml of hydrazine hydrate and heated at reflux for 1 h. The reaction mixture was then concentrated under vacuum, partitioned between 5% aqueous NaHCO3 and methylene chloride and dried over magnesium sulfate. To the methylene chloride extracts were added 12.5 g of 4-trifluoromethylphenyl isocyanate and the mixture was then refluxed one hour, cooled to room temperature and concentrated to 26.1 g of a brown oil. Chromatography on silica gel followed by trituration with ether afforded 11.78 g of the title compound as a tan powder, m.p. 149-151°C. 'H NMR (CDCl₃), δ 1.9 (m.1H), 2.4 (m,1H), 3.0 (m.2H), 3.5 (m.2H), 4.4 (m.1H), 7.3 (m.3H), 7.58 (d,2H), 7.65 (d,2H), 8.0 (d,2H), 8.3 (bs,1H).

Example 2

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3,3a.4,5-tetrahydro-3a-methyl-N-[4-(trifluoromethyl)phenyl]-2H-benz[g]indazole-2-carboxamide

To a solution of 0.9 ml of diisopropylamine in 10 ml of THF, at -78°C, was added 2.3 ml of 2.5 M n-butyl lithium in hexane and the mixture was stirred for 5 mins. To this -78°C solution was added a solution of 1.0 g of the title compound of Example 1 in 5 ml of THF. The reaction was warmed to 0°C, recooled to -78°C and then 0.4 ml of methyl iodide was added. The reaction mixture was then stirred for 24 h, with gradual warming to room temperature, quenched with 0.5 ml of glacial acetic acid, and poured into a 5% solution of aqueous NaHCO₃. The mixture was extracted with chloroform, dried over magnesium sulfate and concentrated. Crystallization from 1:1 ethylacetate-hexane afforded 0.41 g of a yellow solid, m.p. 130° to 135°C. Chromatography of the filtrate on silica gel with 30% ethyl acetate hexane afforded an additional 0.17 g of a higher purity sample of the title compound, m.p. 152.5° to 153.5°C. 'H NMR (CDCl₃), δ 1.29 (s, 3H), 2.1 (m.2H), 2.9-3.3 (m. 2H), 3.70 (d, 1H), 4.12 (d,1H), 7.3 (m,3H), 7.57 (d,2H), 7.68 (d,2H), 7.98 (d,1H), 8.3 (bs.1H).

Example 3

$\underline{\text{Methyl}} \hspace{0.1cm} \textbf{3.3a.4.5-tetrahydro-2-[[4-(trifluoromethyl)phenylamino]carbonyl]-2H-benz[g]indazole-3a-carboxylate}$

Application of the procedure of Example 2 with 0.75 ml of methyl chloroformate afforded, after chromatography on silica gel. 0.25 g of the title compound as a white solid, m.p. 177° to 180°C. H NMR $(CDCl_3)$, δ 2.1 (m.1H), 2.7 (m.1H), 3.0 (m,2H), 3.71 (s.3H), 3.76 (d.1H), 4.58 (d.1H), 7.3 (m.3H), 7.57 (d.2H), 7.66 (d.2H), 8.0 (d.1H), 8.22 (s.1H).

Example 4

7-Chloro-3.3a,4.5-tetrahydro-N-[4-(trifluoromethyl)phenyl]-2H-benz[g]indazole-2-carboxamide

Step A: ((2-(3-chlorophenyl)ethyl)methanesulfonate

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To a 0°C solution of 30.0 g of 3-chlorophenethyl alcohol and 15.3 ml of methane sulfonyl chloride in 150 ml of THF was added, dropwise, a solution of 28.0 ml of triethylamine in 50 ml of THF. The reaction was warmed to room temperature, stirred overnight, and then filtered. The filtrate was partitioned between aqueous sodium bicarbonate and ether. The organic extracts were then dried over magnesium sulfate, filtered and concentrated under reduced pressure to afford 45.93 g of a clear, colorless oil. 'H NMR was consistent with the structure for Step A.

30 Step B: 3-chlorobenzenebutanoic acid

To a mixture of 8.0 g of 60% sodium hydride in 300 ml of THF, under N_2 , was added dropwise a solution of 31.0 ml of diethyl malonate in 50 ml of THF. Upon complete addition of the diethyl malonate, a pale yellow homogenous solution was obtained. To this was added a solution of 45.93 g of the sulfonate from Step A and the mixture was then heated at reflux overnight. The reaction was then cooled to room temperature, poured into 400 ml of 1N HCl, and extracted with ether. The ether extracts were dried over magnesium sulfate, filtered and concentrated under reduced pressure to afford 68.9 g of a yellow oil. The crude oil was dissolved in 400 ml of methanol, 100 ml of H_2O and 40 ml of 50% aqueous NaOH. The reaction was stirred overnight and the methanol was then removed at reduced pressure. The crude residue was partitioned between H_2O and ether, the aqueous extracts were acidified with concentrated HCl and then extracted several times with ether. The ether extracts were dried over magnesium sulfate, filtered and concentrated to afford 51.7 g of a yellow oil. The crude residue was dissolved in 200 ml of toluene and heated at reflux for 4 days under N_2 to effect decarboxylation. After this time toluene was removed by concentration at reduced pressure to afford 35.72 g of a yellow oil. 'H NMR analysis of the crude product was consistent with 3-chlorobenzenebutanoic acid of purity estimated to be 80%. The crude product was used without further purification directly in the next step.

50 Step C: 6-chloro-3,4-dihydro-1(2H)-naphthalenone

A mixture of 35.72 of the product from Step B and 50 ml of thionyl chloride was heated at reflux for 2 hours and then stirred at room temperature for 18 hrs. After this time thionyl chloride was removed at reduced pressure and the product was dissolved in carbon tetrachloride and concentrated at reduced pressure. The residue was dissolved in 150 ml of dichloroethane cooled to 0°C and 28 g of aluminum chloride was added portionwise over about 1 hr in approximately 3 g portions. After stirring for 3 hrs the

reaction was poured over a mixture of ice-1N HCl and extracted three times with methylene chloride. The organic extracts were dried over magnesium suifate and concentrated to approximately 30 g of a brown oil. Chromatography on silica gel with 10% ethyl acetate hexane afforded 17.83 g of 6-chloro-3.4-dihydro-1(2H)-naphthalenone as a brown oil. H NMR was consistent with the structure.

Step D: 7-chloro-3,3a,4,5-letrahydro-N-[4-(trifluoromethyl)phenyl]-2H-benz[g]indazole-2-carboxamide

A mixture of 6-chloro-3,4-dihydro-1(2H)-naphthalenone (Step C), 2.5 g of dimethylamine hydrochloride. 1.0 g of paraformaldehyde, 0.7 ml of concentrated HCl and 15 ml of ethanol was combined and heated at reflux for 18 hrs. The reaction was then concentrated at reduced pressure and partitioned between H₂O and ether. The aqueous extracts were made basic with 1N NaOH and then extracted three times with ether. The ether extracts were dried over magnesium sulfate and concentrated to 4.64 g of a yellow oil. This compound was dissolved in 25 ml of ethanol and 1.5 ml of hydrazine hydrate was added followed by 5 to 6 drops of 50% sodium hydroxide. The reaction was then heated at reflux, under N₂, for 2 to 3 hrs after which time it was cooled and most of the ethanol was removed by concentration at reduced pressure. The crude residue was partitioned between saturated aqueous sodium bicarbonate and methylene chloride. The methylene chloride extracts were dried over magnesium sulfate and filtered. The methylene chloride extracts were then combined with 3.5 g of 4-trifluoromethylphenyl isocyanate and stirred under N₂ overnight. The reaction was then concentrated and the crude residue triturated with ether to afford 3.35 g of the title compound as a white powder, m.p. 196 to 199°C.

'H NMR (CDCl₃) δ 1.9 (m. 1H), 2.2 (m. 1H), 3.0 (m. 2H), 3.5 (m. 2H), 4.43 (m. 1H), 7.24 (m. 2H), 7.55 (d.

2H), 7.67 (d, 2H), 7.92 (d, 1H), 8.20 (s, 1H).

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Example 5

Methyl 7-chloro-3.3a,4.5-tetrahydro-2-[[4-(trifluoromethyl)phenylamino]carbonyl]-2H-benz[g]indazole-3a-car-boxylate

A solution of 50 ml of THF and 6.7 ml of diisopropylamine was cocled under N₂ to -78°C and then 17.5 ml of 2.5 M n-butyllithium in hexane was added. After 5 min, a solution of 7.8 g of the title compound of Example 4 in 15 ml of THF was added dropwise and the dark red solution that formed was stirred at -78°C for an additional 15 min. After this time a solution of 4.6 ml of methyl chloroformate in 10 ml of THF was added dropwise and the red color dissipated rapidly. The reaction was warmed to room temperature and after 1 hr quenched with 5% aqueous sodium bicarbonate. The reaction mixture was partitioned between ether and 5% aqueous sodium bicarbonate. The ether extracts were dried over magnesium sulfate and concentrated to 14.1 g of a yellow oily solid. The crude product was triturated with ether and the resulting white precipitate was filtered and dried to afford 5.56 g of the title compound as a white solid, m.p 234 to 236°C. ¹H NMR(CDCl₃) δ 2.1 (m, 1H), 2.75 (m, 1H), 2.95 (m, 2H), 3.71 (s, 3H), 3.75 (d, 1H, J = 6Hz), 4.59 (d, 1H, J = 6Hz), 7.25 (m, 2H), 7.57 (d, 2H), 7.66 (d, 2H), 7.94 (d, 1H), 8.18 (d, 1H).

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Example 6

7-Chloro-2,3,3a,4-tetrahydro-N-[4-(trifluoromethyl)phenyl]-[1]benzothiopyrano[4,3-c]pyrazole-2-carboxamide

Step A: 7-chloro-3-[(dimethylamino)methyl]-2,3-dihydro-4H-1-benzothiopyran-4-one

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A mixture of 15.0 g (75.7 mmol) of 7-chlorothiochroman-4-one (Chu et al., <u>Hua Hsüeh Hsüeh Pao.</u> 1956. 22, 371-8; <u>Chem. Abstracts</u>, 1958, <u>52</u>, 11044a), 7.41 g (90.8 mmol) of dimethylamine hydrochloride, 3.41 g (113 mmol) of paraformaldehyde, and 1.26 ml (15 mmol) of concentrated HCl in 25 ml of ethanol under a

nitrogen atmosphere was heated under reflux for 22 hrs. The reaction mixture was poured into 150 ml of water and extracted with ether (2 \times 50 ml). The aqueous layer was made basic (pH 10) with 1N NaOH and then was extracted with ethyl acetate (3 \times 100 ml). After being dried (MgSO₄), these ethyl acetate extracts were concentrated under reduced pressure to give 14.9 g of a yellow solid. The 'H NMR was consistent with the structure for the title compound.

Step B: 7-chloro-2,3,3a,4-tetrahydro-N-[4-(trifluoro methyl)phenyl]-[1]benzothiopyrano[4,3-c] oyrazole-2-car-boxamide.

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To a stirred mixture of 14.0 g (54.7 mmol) of the Mannich base from Step A and 6.19 ml (6.38 g, 110 mmol) of hydrazine hydrate in 36 ml of n-propanol under a nitrogen atmosphere was added 0.29 ml (5.5 mmol) of 50% aqueous NaOH and the resulting mixture was heated at 100°C for 2 hrs. The reaction mixture was concentrated under reduced pressure, diluted with 250 ml of dichloromethane, and washed with 200 ml of dilute aqueous sodium bicarbonate. The aqueous layer was extracted with 50 ml of dichloromethane. The combined organic layers were dried (MgSO₄) and filtered to give 370 ml of solution. A 120 ml portion of this solution was treated under a nitrogen atmosphere with 3.75 g (20.0 mmol) of 4-(trifluoromethyl)phenyl isocyanate and allowed to stir overnight. The reaction mixture was concentrated under reduced pressure, and the residue was triturated with 35 ml of ether and filtered to give 4.51 g (63%) of the title compound as a yellow solid melting at 188 to 192°C. H NMR (CDCl₃, 200 MHz) δ 3.19 (dd, 4.5Hz, 1H), 3.32 (t, 12.0Hz, 1H), 3.69 (t, 10.8Hz, 1H) 3.86 (dq, 4.5Hz, 11Hz, 1H), 4.43 (t, 10.5Hz, 1H), 7.13 (dd, 1.9Hz, 8.8Hz, 1H), 7.24 (d, 1.9Hz, 1H), 7.58 (d, 8.8Hz, 2H), 7.66 (d, 8.8Hz, 2H), 7.93 (d, 8.4Hz, 1H), 8.13 (s, 1H).

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Example 7

7-chloro-2.3.3a.4-tetrahydro-4,4-dimethyl-N-[4-trifluoromethyl)phenyl]-[1]benzothiopyrano[4.3-c]pyrazole-2-carboxamide

Step A: 3-(3-chlorophenylthio)-3-methylbutanoic acid

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A mixture of 20.0 g (138 mmol) of 3-chlorothiophenol, 13.85 g (138 mmol) of 3-methyl-2-butanoic acid, and 4.1 ml (3.5 g, 42 mmol) of piperidine under a nitrogen atmosphere was heated under reflux at 105° C for 28 hrs. The reaction mixture was poured into 200 ml of ether and washed with 1N HCl (2 × 80 ml) and then the aqueous layers were extracted with 50 ml of ether. The combined organic layers were dried (MgSO₄) and concentrated under reduced pressure to give 33.1 g of a yellow oil. The 'H NMR was consistent with the structure for the title compound.

Step B: 7-chloro-2,3-dihydro-2,2-dimethyl-4H-1-benzothiopyran-4-one

To 270 g of polyphosphoric acid heated to 70°C was added, portionwise, 31.9 g of the product from Step A over 15 min with mechanical stirring under a nitrogen atmosphere. After heating at 70°C for an additional 45 min, the reaction mixture was cooled to 0°C and 150 ml of water was carefully added followed by 100 ml of ether. The resulting mixture was further diluted with 170 ml of water and 110 ml of ether. The layers were separated and the aqueous layer was extracted with ether (2 x 200 ml). The combined ether layers were washed with 100 ml of 1N NaOH and were then dried (MgSO₄) and concentrated under reduced pressure to give 25.8 g of a yellow solid. The 'H NMR was consistent with a 68:32 mixture of the structure for Step B and its 5-chloro isomer, respectively.

Step C: 7-chloro-3-(dimethylamino)methyl]-2,3-dihydro-2,2-dimethyl-4H-1-benzothiopyran-4-one

A mixture of 10.0 g (44.1 mmol) of the isomeric thiochromanones from Step B. 1.99 g (66.2 mmol) of paraformaldehyde, 4.32 g (52.9 mmol) of dimethylamine hydrochloride, and 0.74 ml of concentrated HCl in 15 ml of ethanol under a nitrogen atmosphere was heated under reflux for 49 hours. The reaction mixture was poured into 1000 ml of water and extracted with ether (5 x 200 ml). The aqueous layer was made basic (pH 10) with 1N NaOH and then extracted with ethyl acetate (3 x 200 ml). After being dried MgSO₂), these ethyl acetate extracts were concentrated under reduced pressure to give 3.59 g of an oii. The "H NMR was consistent with a 80:20 mixture of the structure for the title compound of Step C and its 5-chloro isomer, respectively.

Step D: 7-chloro-2,3,3a,4-tetrahydro-4,4-dimethyl-N-[4-(trifluoromethyl)phenyl]-[1]benzothiopyrano[4,3-c]pyrazole-2-carboxamide

To a stirred mixture of 3.20 g (11.3 mmol) of the Mannich bases from Step C and 1.28 ml (1.32 g, 22.6 mmol) of hydrazine hydrate in 7.5 ml of n-propanol under a nitrogen atmosphere was added 90 mg (1.1 mmol) of 50% aqueous NaOH and the resulting mixture was heated at 100°C for 2 hrs. The reaction mixture was concentrated under reduced pressure, diluted with 50 ml of dichloromethane, and washed with 100 ml of dilute aqueous sodium bicarbonate. The aqueous layer was extracted with 20 ml of dichloromethane. The combined organic layers were dried (MgSO₄) and filtered to give 93 ml of solution. A 27 ml portion of this solution was treated with 0.68 g (3.6 mmol) of 4-(trifluoromethyl)phenyl isocyanate and allowed to stir at room temperature for 2.5 hrs. The reaction mixture was concentrated under reduced pressure, and the residue was triturated with 15 ml of ether and filtered to give 1.03 g (71%) of the title compound as a yellow solid, m.p. 200° to 202°C. 'H NMR (CDCl₃, 200 MHz) § 1.35 (s. 3H), 1.45 (s. 3H), 3.83 (m. 2H), 4.21 (m. 1H), 7.14 (dd, 2.0 Hz, 8.4Hz, 1H), 7.22 (d, 1.8Hz, 1H), 7.58 (d, 8.8Hz, 2H), 7.66 (d, 8.9Hz, 2H), 7.95 (d, 8.1Hz, 1H), 8.13 (s. 1H).

Example 8

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35 3a-(4-chlorophenyl)-3,3a,4,5-tetrahydro-N-[4-(trifluoromethyl)phenyl]-2H-benz[g]indazole-2-carboxamide

Step A: 4-chloro-2-(2-phenylethyl)benzeneacetic acid, methyl ester

To a solution of 4.35 g of 60% sodium hydride in 50 ml of DMF was added dropwise 20.0 g of methyl 4-chlorophenyl acetate in 25 ml of DMF. Once the evolution had ceased a mixture of 20.0 g of 2-bromoethyl benzene in 25 ml of DMF was added and the reaction was stirred at room temperature overnight. The reaction was then partitioned between 400 ml of ether and 200 ml of H_2O . The ether extracts were washed with H_2O , dried over magnesium sulfate, filtered and concentrated to 32.9 g of a yellow oil. Chromatography on silica gel with 5% ethyl acetate in hexane afforded 12.6 g of the title compound as a clear, colorless oil.

Step B: 2-(4-chlorophenyl)-3,4-dihydro-1(2H)-naphthalenone

A mixture of 12.6 g of the title compound from Step A, 100 ml of methanol, 30 ml of H_2O and 10 ml of 50% NaOH were refluxed under N_2 overnight. The reaction was then concentrated at reduced pressure and concentrated HCl was added until acidic. The mixture was extracted three times with ether, dried over magnesium sulfate and concentrated to afford 12.14 g of a yellow oil. The residue was dissolved in 20 ml of thionyl chloride and heated at reflux under N_2 , overnight. Thionyl chloride was removed by concentration at reduced pressure and the residue was concentrated twice from carbon tetrachloride. The crude product was dissolved in 50 ml of dichloroethane, cooled under N_2 to 0°C and 5.9 g of aluminum trichloride was added.

The reaction was stirred overnight at room temperature, poured into 50 ml of 1N HCl and extracted with methylene chloride. The organic extracts were dried over magnesium sulfate, filtered and concentrated to 12.3 g of a brown solid. Chromatography on silica get with 10% ethyl acetate hexane followed by trituration with methanol afforded 5.51 g of the title compound as a brown solid, m.p. 97 to 99°C.

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Step C: 2-(4-chlorophenyl)-3,4-dihydro-2-(hydroxymethyl)-1(2H)-naphthalenone

To 10 ml of ethanol was added a solution of 1.0 g of the title compound from Step B in 2 ml of toluene followed by 1.5 ml of 37% formaldehyde and then 7.8 ml of 0.5 M NaOH. The reaction was stirred at room temperature, under N₂, for four hours after which time TLC confirmed disappearance of the starting tetralone. The reaction was poured into a mixture of 100 ml 5% aqueous sodium bicarbonate and 100 ml of ethyl acetate. The aqueous extracts were washed twice with ethyl acetate, the organic extracts were then dried over magnesium sulfate, filtered and concentrated at reduced pressure. Chromatography on silica gel with 20% ethyl acetate/hexane afforded 1.19 g of a brown oil. H NMR was consistent with the structure of the title compound.

Step D: 2-(4-chlorophenyl)-3,4-dihydro-2-[(methylsulfonyloxy)methyl]-1(2H)-naphthalenone

To a mixture of 0.99 g of the title compound of Step C and 0.6 ml of methanesulfonyl chloride in 20 ml of THF at 0°C was added 1.05 ml of triethylamine. The reaction was stirred overnight at room temperature, partitioned between 5% aqueous sodium bicarbonate and chloroform, dried over magnesium sulfate and concentrated. The residue was chromatographed on silica gel with 30% ethyl acetate hexane to afford 1.1 g of the title compound as a viscous yellow oil.

30 <u>Step</u> E: 3a-(4-chlorophenyl)-3.3a,4,5-tetrahydro-N-[4-(trifluoromethyl)phenyl]-2H-benz[g]indazole-2-carbox-amide

A mixture of 0.9 g of the title compound of Step D. 10 ml of ethanol and 0.4 ml of hydrazine hydrate was refluxed under N₂ for 4 days. The reaction was cooled to room temperature and the white precipitate which formed was filtered. The filtrate was partitioned between saturated aqueous potassium bicarbonate and ethyl acetate. The organic extracts were dried over magnesium sulfate, filtered and concentrated at reduced pressure to afford 0.58 g of a gummy yellow solid. The crude residue was dissolved in 5 ml of THF and 0.4 ml of 4-trifluoromethylphenyl isocyanate was added. After 15 minutes TLC on silica gel (30% ethyl acetate/hexane) indicated formation of a fluorescent product. Concentration of the reaction mixture under reduced pressure and chromatography on silica gel with 20% ethyl acetate/hexane followed by chromatography on silica gel with 10% methylene chloride in n-butylchloride afforded after trituration with hexane 49.2 mg of a beige powder. ¹H NMR was consistent with the structure of the title compound. ¹H NMR (CDCl₃) w 2.30 (dt, 1H), 2.5-2.9 (m, 3H), 3.90 (d, 1H), 4.33 (d, 1H), 7.14 (m, 3H), 7.2-7.4 (m, 3H), 7.56 (d, 2H), 8.11 (m, 1H), 8.28 (bs, 1H).

By the general procedures described herein, or obvious modifications thereof, the compounds of Tables 1 through 18 can be prepared.

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General Structures for Tables 1-18

5	Q-C(X)NH-		, ,
Tabl	<u> </u>		1
10			
1	Q-1	(J=K=H.	R ₈ =H)
2	Q-2	(J=K=H,	R ₈ =H)
75	Q-3	(J=K=H,	R ₈ =H)
4	Q-4	(J=K=H,	$x=0, x^1=0)$
5	Q-5	(J=K=H.	$x=0, x^1=0)$

Q-1 (J=K=H) Q-2 (J=K=H) Q-3 (J=K=H) Q-4 (J=K=H, $X^{1}=0$) Q-5 (J=K=H, X1=0)) Q-C(0)NH-Q-1 Q-2 Q-3 Q-4 (X1=0) Q-5 (X1=0)

18 Q is R₂

For the tables that follow, the values of Q will be numbered as shown:

For Tables 16, 17 and 18 only one of the two possible enantiomers of Q is shown for clarity. For compounds in Tables 17 and 18, when V is SO, the a isomer represents compounds where the B substituent is cis to the oxygen of the sulfoxide. The ß isomer represents those compounds where the B substituent is trans to the oxygen of the sulfoxide.

TABLE 1

5	<u> 2</u>	2 ^b	<u> </u>	Þ	¥	Ŧ	Phys. Prop. (m.p. °C)
	H	H	н	H	0	0	
	4-C1	н	H	H	0	0	
	4-F	H	H	H	0	0	
10	4-Br	H	H	H	0	Ö	
	4-CF ₃	H	H	н			
	4-OCF ₂ H	H	H	H	0	0	
15	4-CF ₃	3-C1	H	H	0	0	
/3	4-CO ₂ He	H	H	H	0	0	
	4-co ₂ -i-Pr	H	H	н	0	0	
	4-802	н	H	н	0	0	
20	4-SMe	н	H	H	0	0	
	4-SO ₂ He	H	H	Н	0	0	
	4-CF ₂ C1	H	H	H	٥	0	
25	4-0CF ₃	H	H	н	0	٥.	
25	4-I	H	H	H	0	0	
	4-OCF ₂ CF ₂ H	H	H	H	0	0	
	4-CN	H	H	H	0	0	
	4-He	H	H	H	0	0	
30	3,4-CF ₂ CF ₂ O		H	H	0	0	
	3,4-0CF ₂ CF ₂		н	н	0	0	
	3,4-(He)2CC	420	н	H	0	0	
35	4-C1	H	н	Me	0	0	
	4-Br	H	H	Нe	0	0	
	4-CF ₃	H	H	He	0	0	
	4-SMe	H	н	Мe	0	0	
	4-OCF ₂ H	H	H	Нe	0	0	
40	4-CF ₂ Cl	н	н	He	0	0	
	4-co ₂ i-Pr	н	н	Жe	0	0	

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TABLE 1 (CONTINUED)

		_					Phys.
5	R.	$\frac{R_1^b}{H}$	R,	78	×	ţ	Prop. (m.p. °C)
3	4-C1	_ ± H	$\frac{R_2}{H}$	B allyl	o X	<u>t</u>	
	4-Br	H	H	allyl	0	٥	
	4-CF ₃	H	H	allyl	0	0	
		н	н	CH ₂ Ph	0	0	
10	4-C1	н	н	CH ₂ Ph	0	0	
	4-Br				_	0	
	4-CF ₃	н	Н	CH ₂ Ph	0		
	4-CF ₃	3-C1	H	4-F-CH ₂ Ph	0	0	
15	4-SMe	н	H	4-F-CH ₂ Ph	0	0	
	4-C1	н	H	4-F-CH ₂ Ph	0	0	
	4-F	H	H	4-C1-CH ₂ Ph	0	0	
20	4-802	н	H	4-C1-CH ₂ Ph	0	0	
	4-0CF ₂ H	н	н	Ph	0	0	
	4-0CF ₃	н	н	Ph	0	0	
	4-CF ₃	н	н	Ph	0	0-	
25	4-C1	3-01	н	4-C1-Ph	O	0	
	4-Br	H	H	4-Cl-Ph	0	0	
	4-OMe	H	H	4-C1-Ph	0	0	
	4-CF ₃	H	H	n-Bu	0	0	
30	4-C1	н	H	n-Bu	0	0	
	4-CF ₃	H	н	(CH ₂) ₃ Cl	0	0	
	4-Cl	н	н	(CH ₂) ₃ C1	0	0	
	4-Br	H	н	CH ₂ CF ₃	0	0	
35	4-0CF ₃	H	н	CH ₂ CF ₃	0	0	
	4-SCF_H	н	н	CH ₂ OHe	0	0	

TABLE 1 (CONTINUED)

5	Ra	R ₁ b	R.				Phys.
	R1 4-CF3	H	R ₂ H	E CH ₂ OEt CH_CO_Me	<u>x</u> 0	<u>t</u> 0	Prop.
	4-C1	H	H	CH ₂ CO ₂ Me	0	0	
10	4-Br	н	H	CH ₂ CO ₂ Me	0	0	
	4-OCF ₂ H	H	H	CH ₂ CN ²	0	0	
	4-ocf ₂ cf ₂ h	H	H	(CH ₂) ₂ CN	0	0	
15	4-CF ₃	H	H	CO ₂ Me	0	0	
, ,	4-C1	H	H	CO ₂ Me	0	0	
	4-Br	H	H	СО <mark>2</mark> ме	0	0	
	4-F	H	H	CO ₂ Me	0	0	
20	4-CF ₂ C1	H	H	CO ₂ Me	. 0	0	
	4-0CF ₃	H	H	CO ₂ Me	0	0	
	3.4-CF ₂ CF ₂ O		Н	CO ₂ Me	0	0	
25	3.4-OCF ₂ CF ₂		H	CO ₂ Me	0	0	
	3.4-(Me) ₂ CCH ₂	,0	H	CO _Z Me	. 0	0	
	4-CF ₃	н	H	CO ₂ Et	0	0	
30	4-Br	H	H	COZEt	0	0	
	4-CF ₃	H	H	CO ² H	0	0	
	4-C1	H	н	CO ² H	0	0	
	4-CF ₃	H	H	CO ₂ Ph	0	0	
35	4-F	H	H	CO ₂ Ph	0	0	
	4-CF ₃	H	H	CO2CH2CF3	0	0	
	4-CF ₃	H	H	CO2CH2CF3	0	0	
40	4-CF ₃	H	H	CO2CH2CN	0	0	
	4-CF ₃	Н	H	CO ₂ CH ₂ CN	0	0	
	4-CF ₃	н	H	CO2n-C6H13	0	0	

* as a representative example this substitution pattern designates

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TABLE 1 (CONTINUED)

		_					Phys.
_		R.D	I.	•	¥	ŧ	Prop. (m.p. °C)
5	<u> </u>	$\frac{\mathbf{R_1^b}}{\mathbf{H}}$	H ₂	CO allyl	<u>o</u> X	<u>t</u> 0	
	A-CF ₃	n.		E CO_callyl		0	
	4-CF ₃	H	H	CO2CH2Ph	0		
	4-CF ₃	H	H	CO2CH2SHe	0	0	
10	•	н	H	C(0) NH2	0	0	
	4-CF ₃		н	C(0)MH ₂	0	٥	
	4-Cl	H				0	
	4-Br	H	H	C(O) WHHe	0	Ö	
15	4-OCF ₃	H	H	C(0) NHHe	U		
	4-CF ₃	н	H	C(O)NHe2	0	0	
	4-01	н	H	C(0) NMe 2	0	0	
	4-Br	н	н	C(O)NHEt	0	0	
20	4-0CF ₃	H	H	C(S) HH2	0	0	
	4-CF ₃	н	H	C(S)MHHe	0	0	
	4-Cl	н	H	C(S)NHe2	0	0	
	4-Br	H	H	C(S)He	0	0	
25	4-OCF ₃	H	H	C(S)Et	0	0	
	4-CF ₃	н	H	C(S) WHe 2	0	0	
	4-C1	H	H	C(S)NHHe	0	0	
3.4	4-0CF ₃	H	H	OH	0	0	
30	4-CF ₃	н	н	он	0	0	
	4-C1	н	н	OH	0	0	
	4-Br	H	H	OH	0	0	
3.5	4-OCF ₂ H	H .	H	OH	0	0	
35	4-00F ₃	н	н	OMe	0	0	
	4-CF ₃	н	н	OHe	0	0	
	4-C1	н	н	OHe	0	0	
40	7-42						

TABLE 1 (CONTINUED)

5	54	_ b	-				Phys.
•	R1 4-Br	$\frac{R_1^b}{H}$	R ₂	B	X	<u>t</u> 0	Prop. (m.p. °C)
	4-Br	H	H	0-allyl	0		
	4-OCF ₂ H	H	H	0-allyl	0	0	
10	4-0CF ₃	н	H	O-allyl	0	0	
.0	4-CF ₃	H	H	OAc	0	0	
	4-C1	H	H	OAc	0	0	
	4-Br	H	H	OCONHHe	0	0	
	4-OCF ₂ H	H	H	OCONHe ₂	0	0	
15	4-0CF3	H	н	OCO ₂ He	0	0	
	4-CF ₃	н	H	oco ₂ Ph	0	0	
	4-C1	н	Н	OSO ₂ Me	0	0	
20	4-Bc	H	H	oso ₂ Me	0	0	•
	4-0CF ₂ H	H	H	осн ₂ рћ	0	0	
	4-0CF3	Н	H	och ₂ Ph	0	0	
25	H	H	H	н	0	1	
20	4-C1	н	Н	н	0	1	91.5 to 96
	4-F	H	H	н	0	1	
	4-Br	н	H	н	•	1	151 to 152
	4-CF ₃	н	H	н	0	1	149 to 151
30	4-OCF ₂ H	н	н	н	٥	1	
	4-CF ₃	3-C1	н	н	0	1	
	4-002He	Н	Н	н	0	1	
35	4-C0 ₂ -1-Pr	н	Н	н	0	1	
	4-NO ₂	н	н	н	0	1	
	4-SMe	H	H	н	0	1	

TABLE 1 (CONTINUED)

	_8	_ b	_				Phys.
5	<u>R</u> 1	$\frac{R_1^b}{1}$	R ₂	<u>B</u>	×	<u>t</u> 1	Prop. (m.p. °C)
	R1 4-S0 ₂ He	H	H	н	0	1	
	4-CF ₂ C1	H	H	н	0	1	
	4-0CF ₃	H	H	н	٥	1	104 to 107
10	4-I	H	H	н	0	1	
	4-OCF ₂ CF ₂ H	H	H	н	0	1	
		. н	н	н	0	1	
	4-He	H	H	H	0	1	
15	3,4-CF ₂ CF ₂ O		H	н	0	1	
	3,4-0CF ₂ CF ₂		H	н	0	1	
	3,4-(He)2CCH	20	н	н	0	1	
	4-C1	- н	H	Me	0	1	
20	4-Br	H	H	Me	0	1	
	4-CF ₃	н	H	Me	0	1	151.5 to 153.5
	4-SMe	н	H	Me	0	1	
	4-OCF ₂ H	H	H	He	0	1	
25	4-CF ₃	н	н	Et	٥	1	120 to 123
	4-C0 ₂ i-Pr	H	H	He	0	1	
	4-C1	H	н	allyl	0	1	
	4-Br	H	H	allyl	0	1	
30	4-CF ₃	H	H	allyl	0	1	158 to 163
	4-C1	н	H	CH ₂ Ph	0	1	
	4-Br	н	H	CH ₂ Ph	0	1	
35	4-CF ₃	н .	H	CH ₂ Ph	0	1	51 to 55
	4-CF ₃	3-C1	H	4-F-CH ₂ Ph	0	1	
	4-SMe	H	H	4-F-CH ₂ Ph	0	1	
40	4-CF ₃	н	H	4-F-CH ₂ Ph	0	1	
	4-CF ₂	H	H	4-C1-CH_Ph	0	1	gum

45

50

TABLE 1 (CONTINUED)

	R ⁴	рb	Ŕ				Phys.
5	R1 4-NO ₂	$\frac{R_1^b}{H}$	R ₂	<u>B</u> 4-C1-CH ₂ Ph	<u>×</u>	<u>t</u> 1	Prop. (m.p. *C)
	4-0CF ₂ H	н	н	Ph .	0	1	
	4-0CF3	н	н	Ph	0	1	
10	4-CF ₃	н	н	Ph	0	1	
	4-01	3-C1	н	4-C1-Ph	0	1	
	4-BC	H	H	4-C1-Ph	0	1	
	4-0Me	Н	H	4-C1-Ph	0	1	
15	4-CF ₃	н	н	n-Bu	0	1	
	4-C1	Н	H	n-Bu	0	1	
	4-CF ₃	H	H	(CH ₂) ₃ C1	0	1	
	4-C1	н	н	(CH ₂) ₃ C1	0	1	
20	4-Br	н	H	CH ₂ CF ₃	0	1	
	4-0CF ₃	н	H	CH ₂ CF ₃	0	1	
	4-scf ₂ h	Н	H	CH ₂ OMe	0	1	
25	4-CF3	н	H	CH ₂ OEŁ	0	1 .	
	4-C1	H	H	CH2CO2Me	0	1	
	4-Br	н	H	CH2CO2He	0	1	
30	4-OCF ₂ H	н	H	CH ₂ CN	0	1	
30	4-OCF ₂ CF ₂ H	н	H	(CH ₂) ₂ CN	0	1	
	4-CF ₃	Н	H	(CH ₂) ₂ CN	0	1	165 to 169
	4-CF ₃	Н	H	(CH ₂) ₃ CN	0	1	186 to 188
35	4-CF ₃	н .	H	CO2He	0	1	177 to 180
	4-C1	н	H	CO ₂ He	0	1	175 to 177
	4-Br	н	H	CO ₂ He	0	1	175 to 178
40	4-F	н	H	CO ₂ He	0	1	
	4-CF_Cl	Н	H	CO_He	0	1	

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TABLE 1 (CONTINUED)

5	R1 4-00F3	Rb H	R ₂	<u>B</u> co ₂ Me	ž Ž	<u>t</u>	Phys. Prop m.p. (°C)
	3,4-CF ₂ CF ₂ O		Н	CO ₂ Me	0	1	
	3,4-0CF2CF20	•	H	CO ₂ Me	0	1	
10	3,4-(Me) ₂ CCH ₂ O		н	CO ₂ Me	0	1	
	4-CF ₃	H	H	CO ₂ Et	0	1	220 to 222
	4-8r	H	Н	CO ₂ Et	0	1	179 to 182
15	4-C1	H	H	CO ₂ Et	0	1	172 to 174
	4-CF ₃	Н	н	CO ₂ H	0	1	182 to 183
	4-CF ₃	H	H	CO ₂ Na	0	1	230
	4-CF3	H	H	CO ₂ Ph	0	1	
20	4-F	н	н	CO ₂ Ph	O	1	
	4-CF ₃	H	Н	CO_CH_CC13	0	1	194 to 196
	4-CF3	Ħ	н	CO2CH2CF3	0	1	
25	4-CF ₃	H	H	CO (CH ₂)C1	0	1	185 to .188
	4CF3	H	н	CO2(CH)8r	0	1	162 to 167
	4-CF ₃	H	н	CO2CH2CN	0	1	
30	4-CF ₃	H	н	CO2n-C6H13	0	1	
	4-CF ₃	H	н	CO ₂ allyl	0	1	
	4-CF ₃	H	H	CO2CH2Ph	0	1	
35	4-CF3	H	н	CO2CH2SMe	0	1	
	4-CF3	H	н .	C(0)NH2	C	1	250
	4-C1	H	H	C(D)NH2	0	1	
	4-8r	H	н	C(O)NHMe	0	1	
40	4-0CF ₃	H	H	C(0)NHMe	0	1	
	4-CF ₃	H	н	C(0) NMe 2	0	1	226 to 230
	4-C1	H	Н	C(0)NMe ₂	0	1	
45	4-8r	H	H	C(O)NHEL	0	1	
	4-CF ₃	H	H	C(O)NHMe	0	1	220
	4-CF ₃	H	H	C(O)NHallyl	0	1	250
	4-CF ₃	H	H	C(O)N(Me)allyl		1	219 to 221
50	4-CF ₃	Ħ	H	C(O)NHPh-4-C1		1	25 5
	4-cF ₃	H	K	C(O)-piperidin	yl	0	1 235

TABLE 1 (CONTINUED)

	R ₁	₽ ^b	R				Phys.
5	4-ocr	$\frac{R_1^b}{H}$	$\frac{R_2}{H}$	c(s)ин ₂	o X	<u>t</u>	Prop. (m.p. °C)
	4-CF ₃	н	н	C(S) NHHe	0	1	
	4-C1	н	н	C(S)MMe2	0	1	
10	4-Br	н	н	C(S)Me	0	1	
	4-0CF ₃	Н	H	C(S)Et	0	1	
	4-CF ₃	H	H	C(S)NMe ₂	0	1	
	4-01	H	H	C(S)NHMe	0	1	
15	4-0CF ₃	H	H	ОН	ō	ī	
	4-CF ₃	н	H	ОН	0	1	
	4-C1	н	H	ОН	0	1	
	4-Br	H	H	OH	ŏ	i	
20	4-0CF ₂ H	H	H	он	0	1	
	4-0CF3	H	н	OMe	0	1	
	4-CF3	H	Н	OMe	0	1	
25	4-C1	Н	H	OMe	0	1	
	4-Br	H	H	0-ally1	0	1	
	4-OCF ₂ H	H	H	0-ally1	0	1	
	4-OCF ₃	H	H	0-a11y1	0	1	
30	4-CF ₃	н	н	OAc	0	1	
	4-C1	н	н	OAc	0	1	
	4-OCF ₃	Н	Н	OCO ₂ Me	0	1	
	4-CF ₃	н	н	OCO_He	0	1	

TABLE 1 (CONTINUED)

	•	ъ	_				Phys.
5	R ₁	R ₁	R ₂	В	×	<u>t</u>	Prop. (m.p. °C)
J	R1 4-C1	$\frac{R_1^b}{H}$	R ₂	<u>B</u> 050 ₂ He	<u>o</u>	<u>t</u> 1	
	4-Br	н	H	oso ₂ me	٥	,	
	4-OCF ₂ H	н	н	OEL	` 0	T	
10	4-0CF ₃	н	H	OEt	0	1	
	н	H	H	н	0	2	
	4-C1	H	Н	н	0	2	136.5 to 138
	4-F	н	H	H Y	o c	2 2 2	
15	4-Br	н	Н	H	Ĺ	2	
,3	4-CF ₃	н	H	Н	0 ·	2	167 to 168.5
	4-00F ₂ H	Н	н	н	0	2	
	4-CF3	3-C1	H	H	0	2	
20	4-002He	Н	H	н	0	2	
	4-C0 ₂ -i-Pr	H	H	н	C	2	
	4-NO ₂	H	Н	H	0	2 2 2	
25	4-SMe	Н	н	H	0	2	
40	4-50 ₂ Me	H	H	н	0	2	
	4-CF ₂ C1	H	H	н	0	2	
	4-0CF ₃	H	н	н	0	2	
30	4-I	н	H	н	0	2	
	4-OCF ₂ CF ₂ H	H	H	H	0	2	
	4-CN	H	H	н	r	2	
	4-Me	H	H	Н	r	2 2	
35	4-C1	H	H	Me	C	2	137 to 141.5
	4-Br	H	H	He	C	2 2	
	4-CF	H	H	Мe	O	2	134 to 139

TABLE 1 (CONTINUED)

	Ȼ	P p	ъ				Phys.
5	R1 4-SMe	$\frac{R_1^b}{H}$	$\frac{R_2}{H}$	<u>B</u>	<u>×</u>	ţ	Prop. (m.p. °C)
	4-OCF ₂ H	л Н	n H	He He	0	<u>t</u> 2 2	
		Н	н	He	0	2	
	4-CF ₂ C1						
10	4-CO ₂ i-Pr	н	H	He	0	2	
	4-C1	H	н	allyl	0	2 2 2	
	4-Br	Н	H	allyl	0	2	
	4-CF ₃	H	H	allyl	0	2	
15	4-C1	н	H	CH ₂ Ph	0	2	
	4-Br	Н	H	сн ₂ Рћ	0	2	
	4-CF ₃	H	H	CH ₂ Ph	•	2	
	4-CF ₃	3-C1	Н	4-F-CH ₂ Ph	•	2	
20	4-SMe	H	H	4-F-CH ₂ Ph	0	2	
	4-C1	Н	H	4-F-CH ₂ Ph	0	2	
	4-F	н	H	4-C1-CH ₂ Ph	0	2	
25	4-NO ₂	H	H	4-C1-CH ₂ Ph	0	2	
	4-OCF ₂ H	H	H	Ph	0	2	
	4-OCF ₃	H	H	Ph	0	2	
30	4-CF ₃	Н	H	Ph	0	2	
	4-C1	3-C1	H	4-Cl-Ph	0	2	
	4-Br	H	H	4-C1-Ph	0	2 2 2 2	
	4-0Me	H	H	4-C1-Ph	0	2	
	4-CF ₃	H	H	n-Bu	0	2	•
35	4-C1	H ·	Н	n-Bu	٥	2	
	4-CF ₃	H	н	(CH ₂) ₃ C1	ō	2	
	4-C1	н	н	(CH ₂) ₃ C1	0	2	

TABLE 1 (CONTINUED)

5	$\frac{R_1^4}{4-Br}$	$\frac{R_1^b}{H}$	R ₂	E CH ₂ CF ₃ CH ₂ CF ₃	<u>x</u>	<u>t</u> 2	Phys. Prop. (m.p. °C)
				CH ₂ CF ₃			
	4-OCF ₃	н		£ 3	0	2	
	4-SCF ₂ H	H	H	CH ₂ OMe	0	2	
10	4-CF ₃	H	H	CH ₂ OEt	0	2	
	4-C1	H	H	CH ₂ CO ₂ Me	0	2	
	4-Br	н	H	CH ₂ CO ₂ Me	0	2	
15	4-OCF ₂ H	н	H	CH ₂ CN	0	2	
	4-OCF ₂ CF ₂ H	H	H	(сн ₂) ₂ сы	0	2	
	4-CF ₃	н	н	CO ₂ Me	0	2	
	4-C1	н	H	CO2He	0	2	
20	4-Br	н	H	CO2Me	0	2	
	4-F	н	H	CO ₂ Me	0	2	
	4-CF ₂ C1	н	H	CO2He	0	2 .	
25	4-0CF ₃	H	H	CO ₂ Me	0	2	
	4-CF ₃	н	H	COZEt	0	2	
	4-Br	н	н	COZEt	0	2	
30	4-CF ₃	н	н	CO ₂ H	0	2	
	4-C1	H _	н	co ₂ H	0	2	
	4-CF ₃	н	H	CO ₂ Ph	0	2	
J 5	4-F	н	н	CO2Ph	0	2	
Jo	4-CF ₃	н .	н	CO ₂ CH ₂ CF ₃	0	2	
	4-CF ₃	н	H	CO ₂ CH ₂ CF ₃	0	2	

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TABLE 1 (CONTINUED)

	_1	, b	•				Phys.
5	<u>K</u> 1	<u>~1</u>	<u>~2</u>	<u>B</u>	<u>×</u>	<u>t</u>	Prop. (m.p. °C)
-	$\frac{R_1^a}{4-CF_3}$	$\frac{R_1^b}{H}$	ਸ ਮ	CO2CH2CN	0	2	
	4-CF ₃	н	H	CO2CH2CM	0	2	
	4-CF ₃	н	H	CO2n-C6H13	0	2	·
10	4-CF ₃	H	H	CO ₂ mllyl	0	2	
	4-CF3	н	H	CO2CH2Ph	٥	2	
	4-CF ₃	н	H	CO2CH2SHe	0	2	
15	4-CF ₃	н	H	C(0)NH ₂	0	2	
	4-C1	н	н	C(0)NH ₂	0	2	
	4-Br	н	н	C(O)NHHe	0	2	
	4-0CF ₃	н	H	C(O)NHMe	0	2	
20	4-CF ₃	н	н	C(O)NMe2	0	2	
	4-C1	H	н	C(O)NHe2	٥	2	
	4-Br	н	H	C(O) WHEL	0	2	
4-	4-OCF ₃	H	н	C(S)NH ₂	0	2 2 .	
25	4-CF ₃	н	H	C(S)NHMe	0	2	
	4-C1	H	H	C(S)NMe ₂	0	2	
	4-Br	H	H	C(S)He	0	2	
30	4-OCF ₃	Н	н	C(S)Et	٥	2	
	4-CF ₃	н	H	C(S)NMe2	0	2	
	4-Cl	н	H	C(S)NHMe	0	2	
	4-OCF ₃	H	н	ОН	0	2	
35	4-CF ₃	н ,	н	он	٥	2	
	4-C1	H	н	OH	0	2	

TABLE 1 (CONTINUED)

5	R ₁ 4-Br 4-OCF ₂ H	к <u>ъ</u> н н	<u>R</u> 2 Н	он Б	<u>x</u> 0	<u>t</u> 2 2	Phys. Prop. (m.p. °C)
•	4-0CF ₃	H	H	OHe	0	2	
10	4-CF ₃	Н	н	OMe	0	2	
	4-C1 4-Br 4-OCF ₂ H	н н н	н н н	OMe O-allyl O-allyl	0	2 2 2	
	4-OCF ₃	н	н	0-allyl	0	2	
15	4-CF ₃	н	н	OAc	0	2	
	4-C1	н	Н	OAc	0	2	
	4-Br	н	H	OCONHMe	ŏ	2	
20	4-0CF ₂ H	н	H	OCOMHe 2	0	2	
	4-0CF3	н	Н	OCO ₂ Me	0	2	
	4-CF3	Н	н	OCO_He	0	2	
	4-C1	н	H	OSO ₂ He	0	2	
25	4-Br	н	н	OSO ₂ Me	0	2	
	4-0CF ₂ H	н	н	OCH ₂ Ph	0	2	
	4-0CF3	н	н	OCH ₂ Ph	0	2	
30	4-CF ₃	н	5-C1	н	0	0	
	4-0CF3	н	4-C1	Нe	0	0	
	4-0CF ₂ H	н	5-C1	CO ₂ Me	0	0	
35	4-C1	Ĥ	5-C1	CO2CH2CH2C1	0	0	
	4-Br	н	5-C1	н	0	1	164 to 166
	4-CF ₃	H	5-C1	Жe	0	1	
	4-0CF ₃	H	5-C1	Et	0	1	
40	4-C1	н	5-C1	Н	0	1	155 to 157
	4-SMe	H	5-C1	н .	0	1	159 to 161
	4-CO ₂ Et	Н	5-C1	Н	0	1	153 to 156
	4-Br	Н	5-C1	CO ₂ Me	0	1	234 to 236
4 5	4-SHe	н	5-C1	CO ₂ Me	0	1	202 to 204
	4-CO_Bt	н	5-C1	CO_Me	0	1	197 to 199

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TABLE 1 (CONTINUED)

5	R1 4-0CP ₂ H	$\frac{R_1^b}{H}$	R ₂ 5-C1	B CO ₂ He	<u>x</u>	<u>t</u>	Phys. Prop. (m.p. *C)
	4-C1	н	5-C1	CO ₂ He	0	1	222 to 233
10	4-Br 4-CF ₃	H H	5-C1 5-C1	C(0)He C(0)Et	0	1	
	4-00F ₃	н	5-01	C(O)MMe2	0	1	
	4-0CF ₂ H	н	5-C1	OH 2	0	1	
15	4-C1 4-Br 4-CF ₃	н н н	5-C1 5-C1 5-C1	OAc H He	0 0	1 2 2	
	4-0CF ₃	H	5-C1	CO ₂ Me	٥	2	
20	4-0CF2H	н	5-C1	CO ₂ -n-Bu	0	2	
20	4-C1	H	5-C1	CH ₂ Ph	0	2	
	4-Br 4-CR	H H	5-C1	C(O)MHPh	0	2	
	4-CF ₃		5-01	OMe	0	2	
25	4-0CF ₃	H 	5-C1	OC(O)NHMe	0	2	
	4-CF ₃	Н	5-F	CO2He	0	0 .	
	4-0CF ₃	н	5-F	CO ₂ Me	0	0	
30	4-OCF ₂ H	н	5-F	CO ₂ Me	0	0	
	4-CF ₃	H	5-F	н	0	1	175 to 178
	4-C1	H	5-F	н	0	1	
	4-Br	H H	5-F	H	0	1	
35	4-0CF ₂ H		5-F	H	0	1	
	4-C1	н	5-F	CO ₂ Me	O	1	
	4-Br	H ·	5-F	CO ₂ Me	0	1	
	4-CF ₃	н	5-F	CO ₂ Me	0	1	223 to 225
40	4-0CF ₃	H	5- F	CO ₂ Me	0	2	
	4-OCF H	н	5-F	CO He	0	2	

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TABLE 1 (CONTINUED)

	_a	_p b	70				Phys.
5	$\frac{R_1^4}{4-C1}$	$\frac{R_1^b}{H}$	R ₂ 5-P	B CO ₂ Me	o X	<u>t</u> 2	Prop. (m.p. (°C)
	4-Br			CO2Me	0	0	
	4-CF ₃	н	5-0CF ₂ H		0	0	
10	4-0CF ₃	н			0	0	
	4-0CF ₂ H		5-0CF ₂ H	CO2Me	0	1	
	4-C1	н	5-OCF ₂ H	CO2Me	0	1	
15	4-Br	н	5-OCF ₂ H	CO2Me	0	1	
	4-CF ₃	н	5-OCF ₂ H	CO ₂ Me	٥	2	
	4-0CF ₃	н	5-OCF ₂ H	CO2Me	0	2	
	4-0CF ₂ H	н		CO2He	0	2	
20	4-C1	н	5-Br	CO2He	0	1	
	4-Br	н	5-¥0 ₂	CO2He	0	1	
	4-CF ₃	н	_	CO2He	0	1	
25	4-0CF ₃	н	_	CO2He	0	1.	
	4-CF ₃	н	5-OCH ₂ Ph	Же	0	1	
	4-CF ₃	н	5-C(0)Me	Et	0	1	
30	4-CF3	н	5-OCF2CF2H	CO ₂ He	0	1	
	4-CF3	н	5-I	Me	0	1	
	4-CF3	н	5-0CF ₃	Et.	0	1	
35	4-CF3	Н	5-Et	C(0)He	0	1	
33	4-CF3	H .	5-Me	C(O)CF ₃	0	1	
	4-CF3	н	5-CN	Ke	0	1	
	4-CF3	н	5-0Me	CO ₂ Me	0	1	180 to 186
40	4-CF ₃	н		CO2Et	0	1	160 to 166
	4-CF3	н	5-0Me	H	0	1	158 to 160
	4-01	H	5-0 % e	H	0	1	169 to 171

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TABLE 1 (CONTINUED)

	R ₁ 4-CF ₃	$\frac{R_1^b}{H}$	R ₂ 5-co ₂ Me	<u>в</u> n-с ₆ н ₁₃	<u>×</u>	<u>t</u> 1	Phys. Prop. (m.p.*C)
10	4-CF ₃	н .	5-SCF ₂ H	6 13 C(O)H	0	1	
	4-CF ₃	н	5-502He		0	1	
	4-CF ₃	н	5,6-di-C1		٥	1	229 to 230
15	4-CF ₃	н	5,6-di-Cl	CO2He	0	1	238 to 241
	4-CF ₃	н	4-F	H .	0		183 to 184
	4-C1	н	4-P	CO ₂ Me	0	1	
20	4-Br	H	4-2	CO2Me	0	1	
	4-CF ₃	н	4-C1	н	0		179 to 180
	4-0CF ₂ H	н	4-C1	н	0	1	
25	4-0CF ₂ H	н	4-C1	CO ₂ Me	0	1	
••	4-CF ₃	н	4-Br	н	0	1	
	4-CF ₃	н	4-Br	CO ₂ Me	0	1.	
	4-CF ₃	н	6-Cl	CO2He	0	1	
30	4-CF3	н	6-F	CO2He	0	1	
	4-CF ₃	н	6-CF ₃	CO ₂ He	0	1	
	4-CF ₃	н	4-C1	CO ₂ He	0	1	
35	4-CF ₃	н	4-F	CO2He	0	1	206 to 212
	4-CF ₃	н	6-Br	CO2He	0	1	
	4-CF ₃	н	6-NO ₂	CO ₂ He	0	1	
40	4-CF ₃	н	6-Ke	CO ₂ He	0	1	
-	4-CF ₃	н	4-C1	CO2Me	0	1	209 to 211
	4-CF ₃	н	4-F	CO2He	0	1	
	4-CF ₃	н	4-Br	CO ₂ He	0	1	
4 5	4-CF3	н	Н	Н	S	0	
•	4-C1	н	н	Me	S	0	
	4-CF ₃	н	н	CO ₂ He	5	0	
50	4-C1	н	н	н	S	1	
	4-CF ₃	Н	Н	He	S		
	4-C1	н	Н	CO ₂ Me	S	1	
55	4-CF ₃	н	Н	н	S	2	
) <i>U</i>	4-C1	Н	н	Me	S	2	

TABLE 1 (CONTINUED)

5	$\frac{R_1^a}{4-CF_3}$	$\frac{R_1^b}{H}$	R ₂	<u>B</u> C0 ₂ He	<u>x</u> s	<u>t</u> 2	Phys. Prop. (m.p. °C)
10	4-C1 4-CF ₂	н н	5-C1 5-C1	H He	s s	1	
	4-C1	н	5-C1	CO ₂ He	S	1	
	4-CF ₃	н	5-F	н	S	1	
15	4-C1 4-CF ₃	H H	5-F 5-F	He CO ₂ He	s s	1	

TABLE 2

5		Rª	_R b	P					Phys.
		$\frac{R_1^2}{4-CF_3}$ $4-CF_2C1$	R ^b H	$\frac{R_2}{H}$	<u>B</u> H	o X	<u>t</u>	ŏ	Prop. (m.p. °C)
		4-CF ₂ C1	3-C1	н	H	0	1	٥	
10		4-CF3	н	н	н	0	1	0	161 to 164
		4-C1	3-C1	н	н	0	1	0	220 to 224
		4-Br	H	H	H	0	1	0	215 to 220
15		3-CF ₃	н	H	Н	0	1	0	165 to 167
,,	1	4-C1	H	H	н	0	1	0	192 to 199
		3-C1 4-CF ₃	H H	H 5-Br	H H	0	1	0	162 to 171
									192 to 195
		4-Cl 4-Br	H H	5-Br 5-Br	H H	0	1	0	144 to 150
20		4-C1	H	7-8r	л Н	0	1	0	
		4-CF ₃	н	7-Br	H	ŏ	1 1 1	ŏ	
		4-Br	н	7-Br	н	0	1	0	179 to 202
		4-CF ₃	Н	5-C1	 Н	Ö	i	ŏ	152 to 182
25		4-C1	н	5-C1	Н	0	1	0	193 to 207
		4-C1	н	5-F	н	0	1	0	
		4-C0 ₂ Me	H	5- F	H	0	2	0	
30		4-Br	н	5-C1	н	0	1	0	228 to 232
	,	4-CF ₃	H	7-C1	н	0	1	0	178 to 202
		4-C1	н	7-Cl	н	0	1	0	179 to 193
		4-Br	H	7-C1	н	0	1	0	201 to 208
35		4-CF ₃	н	н	Жe	0	1	0	
30		4-CF ₂ C1	3-C1.	5-C1	Xe	0	2	0	
		4-0CF ₃	н	5-C1	Жe	0	0	0	
40	\$	4-ocf ₂ h	Н	5 -F	Жe	0	1	0	
		4-C1 4-C0 ₂ He	H H	5- F H	Me Me	0	2	0	
		A-Rr	¥	u	Wa .	0	1	0	
		4-CF ₃	H	5-C1	CONE	Ö			
45)	4-CF ₂ C1	3-C1	5-C1	CO ₂ Me	0	0		
		4-0CF ₃	н		CO ₂ Me	0	1		
•		4-OCF ₂ H	H	н	CO ₂ Me		2	0	
50		4-C1		5-C1	CO ₂ Me		0		
		4-CO ₂ He			CO ₂ Me		1		
	5	4-Br	н	5- F		0		0	
55		4-CF ₃	н	н	COZEŁ	0	0	0	
		4-C1	н	н	CO_CH_CF3	0	1	0	

TABLE 2 (CONTINUED)

5	p4	_p b						Phys.
	R ₁ 4-CF ₃	$\frac{R_1^b}{H}$	R ₂ 5-C1	E CH ₂ CF ₃	Š O	<u>t</u> 2	<u>v</u>	Prop. (m.p. °C)
10	4-C1 4-CF ₃	H H	5-01 5-00F ₂ H	C(O)Me He	0	0	0	
	4-CF ₃	н	5-0CF ₂ H	CO ₂ He	0	2	0	
	4-C1 4-CF ₃	H H	5-F 5-F	CONHINE CONHPh	0	0 1	0	
15	4-C1 4-CF ₃	H H	н н	allyl CH ₂ Ph	0	2	0	
	4-C1 4-CF ₃	H.	5-C1 5-C1	n-Pr C(0)(CH ₂) ₃ Cl	0	1 2	0.	
20	4-C1	н	5-0CF ₂ H	Me	0	0	0	•
	4-CF ₃	н	5-0CF_H	CO ₂ He	0	1	0	
	4-C1	н	5- F	CH ₂ CO ₂ He	0	2	0	
25	4-CF ₃	H	5-F	CO2H	0	٥	٥	
	4-C1	H	H	4-C1-Ph	0	1.	٥	
	4-CF ₃	Н	н	4-C1-Ph	0	2	0	
30	4-C1	н	5-C1	CO ₂ He	\$	0	0	
	4-CF ₃	н	5-C1	CO2He	S	1	٥	
	4-Cl	Н	5-0CF ₂ H	CO ₂ Me	s	2	0	
	4-CF ₃	н	H .	н	0	0	S	
35	4-CF ₂ Cl	3-C1	H	н	0	1	S	
	4-0CF ₃	H	5-C1	H	0	2	s	
	4-OCF ₂ H	н	5-Cl	н	0	0	S	
40	4-C1 4-CO ₂ Me	H H	5-F 5-F	H H	0	1 2	s s	
	4-Br	H	H	н	0	٥	S	
	4-CF ₃	H	H	H	0	1	S	170 to 171
45	4-C1 4-Br	3-C1 H	H H	H H	0	1	s s	235 to 239 189 to 192

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TABLE 2 (CONTINUED)

	R1	E,	R ₂	<u>B</u>	×	ţ	¥	Phys. Prop. (m.p. °C)
5	3-CF ₃	H	н	H	0	1	S	203 to 204
	4-C1	н	н	н	0	1	S	178 to 180
	3-CF ₃	Н	5-C1	н	ō	ī	8	197 to 200
10	4-C1	Н	5-C1	н	0	1	S	202 to 206
10	3-Gl 4-Br	H H	5-C1	H	0	1	s	27: to 227
	4-C1	3-C1	5-C1 5-C1	H H	0	1	s s	20/ to 211 220 to 232
	4-CF ₃	H	6-C1	н	ō	î	S	195 to 202
15	3-CF ₃	H	6-C1	н	0	1	S	207 to 213
	4-C1	H	6-C1	н	0	1	S	180 to 188
	3-C1	H	6-C1	Н	0	1	S	230 to 240
	4-Br	H	6-Cl	H 	0	1	S	184 to 189
20	4-C1 4-CF ₃	3-C1 H	6-Cl 5-Br	н н	0	1	S	227 to 237
	3 CF				0		S	212 to 215
	3-CF ₃	H	5-Br	H	0	1	. S	187 to 200
	4-Cl 3-Cl	H H	5-Br 5-Br	н н	0	1	S	192 to 209
25	4-Br	H	5-Br	n H	0	1	S	175 to 190
	4-C1	3-C1	5-Br	H	0	1	s s	188 to 215
	4-CF ₃	н	4-C1	H	0	1	S	187 to 200 207 to 210
	4-C1	Н	4-C1	н	٥	1	S	191 to 193
3 0	4-Br	H	4-01	H	ō	ī	S	188 to 190
	4-CF ₃	H	H	Ме	0	1	S	
	4-CF ₂ Cl	3-C1	5-C1	He	0	2	S	
	4-OCF ₃	Н	5-C1	Me	٥	0	S	
35	4-OCF ₂ H	н	5-F	Me	0	1	S	
	4-C1	H	5-F	He	0	2	S	
	4-C0 ₂ Me	н	н	Me	0	0	5	
40	4-Br	H	H	He	0	1	S	
	4-CF ₃	H	5-Cl	CO ₂ Me	0	2	5	
	4-CF ₂ Cl 4-OCF ₃	3-C1 H	5-C1 H	CO ₂ He		0	S	
45	4-0CF ₂ H			CO Me	0	1	S	
3			 5-c1	CO ₂ Me				
	4-C0 ₂ He			CO ₂ Me			S •	
	4-Br		5-F	С0 ₂ Не С0 ₂ Не		2	\$ \$	
50	4-CF ₃	H	н	COZEt		0	S	•
	3 4-C1		н	CO ₂ CH ₂ CF ₃		1	s	
	4-CF ₃		5-c1	CH ₂ CF ₃		2	s	
55	4-C1	H	5-C1	C(0)He	0	0	s	

	_	_	:	TABLE 2 (CONTI	D)			
5	R1	$\frac{\mathbf{R_1^b}}{\mathbf{I}}$	<u>R</u> 2	B	×	£	¥	Phys. Prop. (m.p. °C)
	4-CF ₃	н	5-0CF ₂ H	Me	0	1	s	
	4-CF3	н	5-0CF ₂ H	CO ₂ Me	0	2	S	
10	4-C1 4-CF ₃	H H	5-F 5-F	CONHINE CONHIPH	0	0	s s	
	4-C1 4-CF ₃	H H	H H	allyl CH ₂ Ph	0	2 0	s s	
15	4-C1 4-CF ₃	H H	5-C1 5-C1	n-Pr c(0)(CH ₂) ₃ c1	0	1 2	s s	
	4-C1	H	5-OCF ₂ H	Me	0	0	S	
	4-CF ₃	н	5-OCF ₂ H	CO ₂ He	0	1	s	
20	4-C1	Н	5-F	CH ₂ CO ₂ He	0	2	s	
	4-CF ₃	H	5-F	CO ₂ H	0	0	s	
	4-C1	н	H	4-C1-Ph	0	1	s	
25	4-CF ₃	H	H	4-C1-Ph	0	2	S	
	4-C1	н	5-C1	CO ₂ Me	S	0	S	
	4-CF ₃	н	5-C1	CO ₂ Me	S	1 .	S	
30	4-C1	н	5-0CF ₂ H	CO ₂ Me	S	2	S	
	4-CF ₃	н	H	н	0	0	MMe	
	4-CF ₂ Cl	3-C1	H	н	0	1	MMe	
	4-0CF ₃	н	5-c1	н	٥	2	MMe	
35	4-OCF ₂ H	H	5-C1	н	0	0	MMe	
	4-C1 4-C0 ₂ He	H H	5-F 5-F	H H	0	1 2	WHe WHe	
40	4-Br	н	н	н	0	0	MMe	
-	4-CF ₃	H	H	Xe	0	1	MMe	
	4-CF ₂ Cl	3-C1	5-01	Me	0	2	MMe	
	4-0CF ₃	н	5-C1	Ke	0	0	MMe	

TABLE 2 (CONTINUED)

5								
	R1 4-OCF ₂ H	$\frac{R_1^b}{H}$	R ₂ 5-F	<u>B</u> He	×	<u>t</u> 1	MW•	Phys. Prop. (m.p. *C)
10	4-C1 4-C0 ₂ He	H H	5-F H	He He	0	2	WHe WHe	
	4-Br 4-CF ₃	H H	H 5-C1	He CO ₂ He	0	1 2	WHe WHe	
15	4-CF ₂ Cl	3-01	5-01	CO ₂ He	0	0	MMe	
	4-0CF ₃	H	Н	CO2He	0	1	Me	
	4-OCF ₂ H	н	н	CO ₂ He	0	2	MMe	
20	4-01	H	5-C1	CO2Me	0	0	MMe	
	4-CO2Me	н	5-C1	CO2He	0	1	WHe	
	4-Br	н	5-F	CO ₂ Me	0	2	MMe	
25	4-CF ₃	H	н	COZEŁ	0	0	MHe	
25	4-C1	н	н	CO2CH2CF3	0	1	Mile	
	4-CF ₃	H	5-C1	CH ₂ CF ₃	0	2	MHe	
	4-C1	н	5-C1	C(O)He	0	٥	WHe	
30	4-CF ₃	н	5-0CF ₂ H	He	0	1	WHe	
	4-CF ₃	н	5-OCF ₂ H	CO ₂ Me	0	2	MHe	
	4-C1 4-CF	H H	5-F	CONHMe	0	0	MHe	
35	4-CF ₃		5-F	CONHPh	0	1	Me	
	4-CF ₃	H H	·H H	allyl CH ₂ Ph	0	2	WHe WHe	
	4-C1	н	5-C1	n-Pr	0	1	WMe	
	4-CF ₃	H	5-C1	C(0)(CH2)3C1	0	2	WHe	
40	4-C1	н	5-OCF ₂ H	Me	0	0	MMe	
	4-CF ₃	Н	5-OCF ₂ H	∞ ₂ He	0	1	WMe	
•	4-C1	H	5-F	CH2CO2He	0	2	WHe	
45	4-CF ₃	н	5-F	со_н	0	0	WHe	
	4-C1	н	н	4-C1-Ph	0	1	MHe	

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TABLE 2 (CONTINUED)

5	<u>R</u> 1	R ^b	<u>R_2</u>	B	×	£	<u>v</u>	Phys. Prop. (m.p. °C)
,								
	4-CF ₃	H	H	4-C1-Ph	0	2	WMe	
	4-C1	H	5-C1	CO ₂ He	S	0	¥Ме	
10	4-CF ₃	н	5-01	CO ₂ Me	S	1	MHe	
	4-C1	Н	5-OCF ₂ H	CO ₂ Me	S	2	MHe	
	4-CF ₃	Н	H	Н	0	1	so ₂	255 to 260
15	4-C1	3-C1	H	н	0	1	so ₂	250 to 255
	4-Br	н	н	н	0	1	so ₂	272 to 274
	3-CF ₃	H	н	н	0	1	so ₂	261 to 263
	4-C1	H	н	Н	0	1	so ₂	247 to 250
20	3-C1	H	H	н	0	1	so ₂	245 to 249
	4-CF ₃	H	5-C1	H	0	1	so ₂	293 to 297
	3-CF ₃	H	5-C1	H	0	1	so ₂ .	217 to 234
25	4-C1	H	5-C1	H	0	1	so ₂	318 to 321
	3-C1	H	5-C1	н	0	1	so ₂	241 to 270
	4-Br	H	5-01	н	0	1	so ₂	290 to 292
30	4-C1	3-C1	5-Cl	н	0	1	so ₂	280 to 285
30	4-CF ₃	н	6-C1	н	0	1	so ₂	290 to 292
	3-CF3	н	6-C1	н	0	1	so ₂	175 to 182
	4-C1	н	6-C1	н	0	1	so ₂	230 to 280
35	3-C1	н	6-C1	н	0	1	so ₂	245 to 269
	4-Br	H	6-C1	н	0	1	so ₂	307 to 310
	4-C1	3-C1	6-C1	н	0	1	so ₂	275 to 278
40	4-CF ₃	H	н	Me	0	0	so ₂	
	4-C1	н	5-C1	Ме	0	1	so ₂	
	4-0CF ₃	H	н	Me	0	2	so	
45	4-CF ₃		5-C1		0	٥	so ₂	
	4-C1				0		so ₂	
	4-0CF ₃	н	5-C1	CO2He			so ₂	
	3-CF3		н		0	1	\$0(a)	255 to 260
50	4-CF		н	н	0	1	\$0(a)	250 to 255

				TABL	.g. 2	(CUN	LINUEDJ	
5	_&	_ b	_					Phys.
3	R ₁	$\frac{R_1^b}{1}$	<u>R₂</u>	<u>p</u>	¥	Ł	¥	Prop. (m.p. °C)
	4-Br	н	н	н	0	1	50(a)	278 to 282
	4-Br	H	H	H	0	1	SO(B)	258 to 260
10	4-C1	3-C1	H	H	0	1	S0(a)	275 to 280
. 0	4-C1	3-C1	Н	Н	0	1	SO(B)	265 to 269
	4-C1	Н	H	H	0	1	\$0(a)	262 to 264
	4-Cl	H	H	H	0	1	SO(B)	247 to 250
	3-C1	H	H	H	0	1	SO(a)	231 to 232
15	3-C1	Н	H	H	٥	1	SO(B)	255 to 258
	4-CF ₃	H	5-C1	Н	0	1	S0(a)	271 to 275
	4-CF3	н	5-C1	Н	0	1	50(5)	256 to 277
	3-CF ₃	H	5-C1	н	0	1	20(a)	210 to 213
20	3-CF ₃	Н	5-C1	Н	0	1	SO(B)	250 to 252
	4-C1	Н	5-C1	H	0	1	20(a)	284 to 286
	4-C1	H	5-C1	H	0	1	SO(B)	278 to 283
	3-C1	H	5-C1	н	0	1	SO(a)	240 to 241
25	3-C1	H	5-C1	Н	0	1	SO(B)	230 to 246
	4-Br	H	5-Cl	H	0	1	\$0(a)	285 to 287
	4-Br 4-Cl	н 3-С1	5-Cl	H H	0	1	20(¥)	267 to 272
	4-C1	3-C1		л Н	0	i	\$0(a)	276 to 280
	4-CI	H 2-C1	6-C1	H .	0	1	50(B) 50(a)	254 to 268 252 to 256
30	4-CF ₃	н	6-C1	н	0	1	SO(B)	218 to 232
	3-CF ₃	н	6-C1	н	0	1	S 0(α)	263 to 265
	3-CF ₃	н	6-C1	Н	0	1	SO(B)	
35				v	_			949 4- 945
	4-Cl 4-Cl	H H	6-C1 6-C1	H H	0	1	50(a)	243 to 245 269 to 272
	3-C1	H	6-C1	H	0	i	SO(B) SO(α)	154 to 160
	3-C1	H	6-C1	Н	Ö	ì	SO(B)	134 60 100
40	4-Br	H	6-C1	н	ō	i	50(a)	248 to 250
40	4-Br	H	6-C1	H	Ö	ī	SO(B)	283 to 285
	4-C1	3-01		H	0	1	\$0(a)	272 to 275
	4-C1	3-C1	6-C1	H	0	1	SO(B)	289 to 295
	ş	H	H	H	0	1	80(8)	229 to 236
4 5	4-CF ₃	H	Н	H	0	1	SO(B)	235 to 258
	4-0CF ₃	H	Н	He	0	0	¥-n-Bu	
	4-CF ₃			CO ₂ Me	0		N-allyl	
50	4-C1	H	H	CO ₂ Me		0	N-Ph	
	4-0CF ₃	H	5-C1	CO ₂ Me	0		M-CH ₂ Ph	
	4-CF ₃	H	H	CO ₂ Me	0	0		
55	4-CF ₃		Н	H	0	1	ИН	
	4-CF ₃	H	н	CO2He	0	1	MH	

	_			TABLE 2 (CONTINUED)						2	
5	$\frac{R_1^a}{}$	R ^b	R ₂	B	×	<u> </u>	_ <u>v</u>				Phys. Prop. (m.p. *C)
	4-CF ₃	н	5-01	н	0	1	K	He			
	4-CF ₃	н	5-c1	CO2X	. 0	1	¥	Нe			
10	_										
				TABLE	2 3						
15	~ a		· _b	_							Phys.
, ,	4-CF ₃		<mark>я</mark> 1 н	<u>в</u> 2 н	<u>B</u> H			<u>×</u>	<u>t</u>	¥ 0	Prop. (m.p. °C)
	4-CF ₂ C1		3-01	н	н			0	1	0	
20	4-0CF ₃		н	5-C1	н			0	2	0	
	4-OCF ₂ H		H	5-C1	H			0	2	0	
	4-C1		H	5-F	H			0		0	
25	4-CO ₂ He 4-Br		н Н	5-F H	н			0	2	0	
20	4-CF ₃		H	H	H He				0	0	
	4-CF ₂ C1		3-C1 ,	5-C1	He			0	2	0	
	4-0CF ₃		н	5-C1	Мe			0	0	0	
30	4-OCF ₂ H		н	5-F	Нe			0	1	0	
	4-01		H	5-F	Хe			0	2	0	
	4-CO ₂ He 4-Br		н н	н н	Me Ma			0	0	0	
35	4-CF ₃		н	5-C1	Me CO ₂ Me			0	.1 2	0	
	4-CF ₂ Cl		3-01	5-C1	CO ₂ Me		(0	0	
	4-0CF ₃		н	н	CO ₂ Me			0	1	0	
40	4-OCF ₂ H		н	н	CO ₂ Me		+	0	2	٥	
	4-C1		н	5-C1	CO2Me		(0	0	0	
	4-C0 ₂ Me		Н	5-CL	CO ₂ Me		(o	1	0	
45	4-Br		Н	5- F	CO ₂ Me		()	2	0	
	4-CF ₃		н .	Н	CO2Et		(>	0	0	
	4-C1		н	Н	CO_CH_C	F	(>	1	0	

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TABLE 3 (CONTINUED)

5	$\frac{R_1^4}{4-CF_3}$	R ₁	R ₂ 5-CL	B CH ₂ CP ₃	× o	<u>t</u> 2	Š Š	Phys. Prop. (m.p. °C)
	4-C1 4-CF ₃	H H	5-01 5-00F ₂ H	C(0)Me Me	0	0	0	
10	4-CF ₃	н	5-0CF ₂ H	CO ₂ Me	0	2	0	
	4-C1 4-CF ₃	H H	5-F	CONHMe CONHPh	0	0	0	
15	4-C1 4-CF ₃	H H	H H	allyl CH ₂ Ph	0	2	0	
	4-C1 4-CF ₃	H H	5-C1 5-C1	n-Pr C(0)(CH ₂) ₃ C1		1 2	0	
	4-C1	н	5-ocf ₂ h	CO ₂ He	0	1	0	
20	4-C1	Н	5-2	CH2CO2He	0	2	0	
	4-CF ₃	н	5-7	CO ₂ H	0	٥	0	
25	4-C1 4-CF ₃	H H	H H	4-C1-Ph 4-C1-Ph	0	1 2	0	
23	4-C1	н	5-C1	CO ₂ Me	5	0	0	
	4-CF ₃	Н	5-C1	CO ₂ Me	S	1	0	
30	4-C1 4-CF ₃	н Н	5-OCF ₂ H H	CO ₂ Me H	s 0	2	o s	
	4-CF ₂ C1	3-C1	н	н	0	1	s	
	4-0CF ₃	Н	5-C1	н	0	2	s	
75	4-OCF ₂ H	н	5-C1	н	0	2	S	
35	4-C1 4-C0 ₂ Me	H . H	5-F 5-F	H H	0	1 2	s s	

5

TABLE 3 (CONTINUED)

	R ₁	_R b	P					Phys.
5	4-Br	$\frac{R_1^b}{H}$	R ₂ H	<u>B</u> H	Ŏ X	ţ	¥ \$	Prop. (m.p. °C)
	4-CF ₃	H	H	He	٥	1	8	
	4-CF2C1	3-C1	5-C1	Же	0	2	S	
10	4-0CF ₃	н	5-C1	Ke	0	0	s	
	4-0CF ₂ H	н	5-F	Me	0	1	s	
	4-C1	н	5-F	Жe	٥	2	S	
	4-CO2He	н	н	He	0	٥	\$	
15	4-Br	н	н	Нe	0	1	s s	
	4-CF ₃	Н	5-C1	CO ₂ He	0	2	S	
	4-CF ₂ Cl	3-C1	5-C1	CO ₂ He	0	0	5	
20	4-0CF ₃	н	H	CO ₂ Me	0	1	S	
20	4-OCF ₂ H	H	H	CO ₂ Me	0	2	S	
	4-C1	H	5-C1	CO ₂ He	0	0	S	
	4-CO ₂ He	H	5-C1	CO2He	0	1	s	
25	4-Br	H	5-7	CO ₂ Me	0	2	8	
	4-CF ₃	н	н	CO ₂ Et	0	0	s	
	4-C1	Н	н	CO2CH2CF3	0	1	S	
30	4-CF ₃	Н	5-C1	CH ₂ CF ₃	0	2	s	
	4-C1	H	5-C1	C(O)He	0	0	S	
	4-CF ₃	Н	5-0CF ₂ H	Мe	0	1	S	
	4-CF ₃	н	5-0CF ₂ H	CO ₂ Me	0	2	s	
35	#_C1	H	5-F	СОИНМе	0	0	s	
	4-CF ₃	Н	5- F	COMHPh	0	1	S	

TABLE 3 (CONTINUED)

5	R ₁ 4-C1 4-CF ₃	ж <u>1</u> н н	ж ₂ н	B allyl CH ₂ Ph	0 Xi O	<u>t</u> 2 0	Y S S	Phys. Prop. (m.p.*C)
10	4-C1 4-CF ₃	H H	5-C1 5-C1	n-Pr C(0)(CH ₂) ₃ C1		2		
	4-C1 4-C1	н н	5-0CF ₂ H 5-F	CO ₂ Me CH ₂ CO ₂ Me	0	1 2	s s	
	4-CF ₃	н	5-F	CO2H	0	0	S	
15	4-C1 4-CF ₃	H H	H H	4-C1-Ph 4-C1-Ph	0	1 2	s s	
	4-C1	н	5-C1	CO ₂ He	s	٥	S	
20	4-CF ₃	н	5-C1	CO2He	S	1	S	
	4-C1 4-CF ₃	H H	5-осғ ₂ н Н	CO ₂ He H	s o	2	S WMe	
	4-CF ₂ Cl	3-C1	H	н	0	1	MMe	
25	4-0CF ₃	H	5-C1	Н	0	2	WMe	
	4-ocf ₂ H	H	5-C1	н	0	2	MMe	
	4-C1 4-C0 ₂ Me	H H	5-F 5-F	H H	0	1 2	V Me V Me	
30	4-Br 4-CF ₃	H H	H H	H He	0	0	MMe MMe	
	4-CF ₂ Cl	3-01	5-C1	Me	٥	2	MMe	
25	4-0CF ₃	H	5-C1	He	0	٥	MMe	
35	4-0CF ₂ H	H	5-F	Me	0	1	MMe	
	4-C1 4-C0 ₂ Me	H H	5- F H	He He	0	2	NMe NMe	

TABLE 3 (CONTINUED)

5	R1 4-Br 4-CF3	<u>в</u> ъ н н	R ₂ H 5-C1	<u>B</u> He	0 X	<u>t</u> 1 2		Phys. Prop. (m.p. °C)
				CO ₂ Ne				
	4-CF ₂ C1		5-C1	CO ₂ Me	0	0	TMe	
10	4-OCF ₃	H	H	CO ₂ Me	0	1	MHe	
	4-ocf ₂ h	H	H	CO ₂ Me	0	2	MMe	
	4-C1	н	5-C1	CO ₂ Me	0	0	MHe	
	4-C0 ₂ He	н	5-C1	CO ₂ Me	0	1	MHe	
15	4-Br	н	5- F	CO2Me	0	2	MMe	
	4-CF ₃	н	н	COZEt	0	0	MMe	
	4-C1	н	н	CO2CH2CF3	0	1	MMe	
20	4-CF ₃	Н	5-C1	CH ₂ CF ₃	O	2	MMe	
	4-C1	н	5-C1	C(0)Me	0	0	MMe	
	4-CF ₃	H	5-00F ₂ H	Нe	0	1	MMe	
	4-CF ₃	H	5-0CF ₂ H	CO ₂ Me	0	ż	MMe	
25	4-C1	н	5- ?	CONHMe	0	0	MMe	
	4-CF ₃	H	5-F	COMMPh	0	1	WHe	
	4-C1	Н	H	allyl	0	2	MMe	
	4-CF ₃	н	н	CH ₂ Ph	0	0	MMe	
30	4-C1	н	5-C1	N-Pr	0	1	MMe	
	4-CF ₃	H	5-Cl	C(0)(CH ₂) ₃ C1	0	2	MMe	
	4-C1	н	5-OCF ₂ H	CO ₂ Me	٥	1	MMe	
35	4-C1	H,	5-F	CH2CO2Me	0	2	MMe	
	4-CF	H	5-F	COH	٥	٥	MM a	

TABLE 3 (CONTINUED)

	R ₁	_R b	Ð					Phys.
5	1 4-C1	$\frac{R_1^b}{H}$	R ₂ H	<u>B</u> 4-C1-Ph	0 X	<u>t</u> 1	Y MMe	Prop. (m.p. °C)
	4-CF ₃	н	Н	4-C1-Ph	0	2	MMe MMe	
	4-C1	н	5-c1	CO2He			MHe	
10	4-CF ₃	н	5-c1	CO2He		1		
	4-C1	н		CO ₂ He			MMe	
	4-CF ₃	н	Н 2	Z Me		0		
	4-C1	н	5-C1	Мe		1	~	
15	4-0CF ₃	н	H	Не			so ₂	
	4-CF ₃	н	5-C1		0	٨	502	
	4-C1	н	н	CO ₂ Me	0			
20	4-0CF ₃	H	5-C1	CO ₂ Me	0		so ₂	
	4-CF ₃	н	н	Me	0		4	
	4-C1	н	5-01	He			50 50	
	4-0CF ₃	н Н	H	He	0	2	S O	
25	4-CF ₃	н	5-C1		0			
	4-61	н	н	CO ₂ He	0		SO	
	4-0CF ₃	н	5-C1		0	2	so	
30	4-CF3	н	Н	Me Z	0	0	SO	
	4-C1	н	5-C1	Me			MH	
	4-0CF ₃	н	н	He	Õ	ō	N-n-B	1
35	4-CF ₃	н .	5-01	CO ₂ Me		•	N-all	
35	4-C1	н	H.	CO ₂ He			N-Ph	•
	4-0CF ₃	н		CO ₂ He			N-CH2	Ph
	4-CF ₃	н	н	CO, Me			¥-4-F-	

TABLE 4

5	$\frac{R_1^a}{4-CF_3}$	$\frac{R_1^b}{H}$	R ₂	<u>в</u> н	<u>×</u>	<u>z</u>	Phys. Prop. (m.p. *C)
	4-C1 4-F 4-OCF ₂ H	н н н	H 5-Cl 5-Cl	H H Me	0 0 0	0 0	
10	4-SHe 4-C1 4-CF ₃	Н 3-С1 Н	Н Н 5-С1	He He Et	0 0 0	0 0 0	
15	4-C1 4-F 4-OCF ₂ H	н н н	5-C1 H H	Et Et allyl	0 0 0	0	
20	4-5Me 4-C1 4-CF ₃	н 3-С1 Н	5-C1 5-C1 5-F	allyl allyl CO ₂ Me	0 0 0	0 0 0	
	4-C1 4-F 4-OCF ₂ H	н н н	5-F 5-0CF ₂ H 5-0CF ₂ H	C(O)Me CH ₂ Ph	0 0	0	
25	4-SHe 4-Cl	H 3-C1	5-CF ₃	CH ₂ Ph Ph Ph	0	0	
30	4-CF ₃ 4-C1 4-F 4-CF ₃	н н н	н н н	Me Me Me H	s s o	0 0 0 Whe	
35	4-C1 4-F 4-OCF ₂ H 4-SMe	н н н	Н 5-C1 5-C1 Н	H H Me Me	0000	nme nme nme	

TABLE 4 (CONTINUED)

5	R ₁ 4-C1 4-CF ₃	<u>R</u> b 3-С1 н	R ₂ H 5-C1	<u>B</u> Me Bt	0 X 0	I NHe NHe	Phys. Prop. (m.p. *C)
10	4-C1 4-F 4-OCF ₂ H	H H H	5-C1 H H	Et allyl	0 0 0	Wide Wide Wide	
	4-SHe 4-C1 4-CF ₃	н 3-с1 н	5-C1 5-C1 5-F	allyl allyl CO ₂ He	0 0 0	Wie Wie Wie	
15	4-C1 4-F 4-OCF ₂ H	н н	5-F 5-OCF ₂ H	C(0)Me CH ₂ Ph	0 0 0	NMe NMe NMe	
20	4-SMe	 Н 3-С1	5-00F ₂ H 5-0F ₃ 5-0F ₃	CH ₂ Ph Ph Ph	0	MHe MHe	
	4-CF ₃ 4-C1	н н	H H	He He	s s	Me Me	
25	4-F 4-CF ₃ 4-C1	н .н .н	н н 5-с1	Me Me	S 0	MMe WH WH	
30	4-Br 4-OCF ₃ 4-CF ₃	н н	5-F H 5-C1	He He He	0 0	NH NH NEt	
	4-Cl 4-Br	H H	5-F H 5-C1	He He	0	NE t NE t	~ W
35	4-OCF ₃ 4-CF ₃ 4-C1	H H	5-F H	He He He	0 0	N-n	CAH ₉

TABLE 4 (CONTINUED)

45	R ₁ 4-Br 4-OCF ₃	<u>н</u> н	R ₂ 5-c1 5-f	<u>B</u> Me Me	<u>×</u>	E N-allyl N-4-Cl-Ph	Phys. Prop. (m.p. °C)
	4-CF ₃	H	H	Нe	0	¥-4-F-Ph	
50	4-C1 4-Br	H H	5-C1 5-F	He Me	0	NPh NCH_Ph	
	4-00F3	н	н	Me	0	NCH_Ph	
	4-CF ₃	н	5-C1	Жe	0	N-CH,C(C1)=CH,	

55

TABLE 5

	_4	~p					Phys.
5	R1 4-CF ₃ 4-OCF ₃	$\frac{R_1^b}{H}$	$\frac{R_2}{H}$	<u>B</u> H	<u>o</u>	z 0	Prop. (m.p. °C)
	4-0CF ₃	н	5-C1	н	٥	0	
	_	Н	н	н	0	0	
10	4-01	Н	5-C1	CO2Ne	0	0	
	4-F	н	н	CO2He	0	٥	
	4-Br	н	5-C1	CO2Me	0	٥	
15	4-CF ₃	н	н	CO2Me	0	0	
٠	4-0CF ₃		5-C1	CO2Me	0	0	•
			н	CO ₂ Me	0	0	
			5-C1	Me	0	0	
20			H	Нe	0		
	4-Br	н	5-C1		0	0	
	4-CF ₃	н	н	Et	ō	Ö	
	4-0CF ₃	Н	5-C1	n-Pr	0	0	
25	4-0CF ₂ H	н	н	allyl	0	ο.	
	4-C1	н	5-C1	allyl	0	0	
	4-F		H		o	ŏ	
30	4-Br	н	5-01	CH2-4-C1-Ph	0	0	
30	4-CF ₃	H		CH ₂ CF ₃	٥	0	
		н	5-C1		0	0	
	4-OCF ₂ H			i-Pr	0	٥	
35			5-C1		0	0	
	4-0CF_	н .	5-C1	OAc	٥	٥	

TABLE 5 (CONTINUED)

	R.	R_1^b	R				Phys.
5	R ₁ ^a 4-OCF ₂ H	3-C1	R ₂ 5-c1	<u>B</u> OMe	<u>×</u>	<u>E</u> 0	Prop. (m.p. °C)
	4-C1	н	5-C1	C(0)CH ₃	0	٥	
10	4-F 4-BC	н н	5-C1 5-C1	C(0) MMe C(S) MMe	0	0	
	4-CF ₃	н	5-F	He	0	0	
	4-0CF ₃	Н	5- F	n-Bu	0	0	
15	4-OCF ₂ H	3-C1	5-P	alkyl	0	0	
13	4-C1 4-F	н н	5-F 5-P	i-Pr CH ₂ Ph	0	0	
	4-Br	H	5-F	Ph	0	0	
20	4-CF ₃	Н	5-CF ₃	CH ₂ CH ₂ F	0	0	
	4-0CF ₃	Н	5-CF ₃	CH ₂ SCH ₃	0	0	
	4-OCF ₂ H	3-C1	5-CF ₃	CH ₂ OCH ₃	0	0	
	4-C1	H	5-CF3	CH ₂ CN	٥	0	
25	4-F	н	5-CF ₃	CH2CO2He	0	٥	
	4-Br	Н	5-CF ₃	CS ₂ He	0	٥	
	4-CF ₃	Н	н	Me	S	0	•
30	4-OCF ₃	H	н	Me	S	0	
	4-OCF ₂ H	3-C1	Н	Me	s	0	
	4-CF ₃	н	Н	н	0	ИМе	
35	4-0CF ₃	Н	5-C1	н	0	MHe	

TABLE 5 (CONTINUED)

	" a	_ b	-				Phys.
5	$\frac{R_1^4}{4-\text{OCF}_2}H$	$\frac{R_1^b}{H}$	R ₂	B H	<u>x</u>	<u>z</u>	Prop. (m.p. °C)
						MMe	
	4-C1	н	5-C1	₂ не	0	MMe	
	4-8	H	H	CO ₂ He	0	MMe	
10	4-Br	H	5-C1	CO ₂ Me	0	MMe	
	4-CF ₃	H	H	CO ₂ Me	0	MMe	
	4-0CF ₃	н	5-C1	CO2Me	0	MMe	
15	4-0CF ₂ H	н	н	CO He	0	MMe	
	4-C1	н	5-C1	Tie	٥	MMe	
	4-F	H	H	Me	0	MMe	_
	4-Br	H	5-C1	Me	0	MMe	•
20	4-CF ₃	н	H	Et	0	MMe	
20	4-OCF ₂ H	Н	Н	allyl	0	MMe	
	4-C1	H	5-C1	allyl	٥	MMe	
	4-F	H	H	CH ₂ -4-F-Ph	0	MMe	
25	4-Br	Н	5-c1	CH2-4-C1-Ph	0	MMe	
	4-CF ₃	H	H	CH ₂ CF ₃	0	NHe	
	4-0CF ₃	Н	5-C1	(CH ₂) ₄ C1	0	₩Me	
30	4-OCF ₂ H	н	н	i-Pr	0	₩He	
30	4-CF ₃	н	5-C1	ОН	0	MMe	
	4-OCF ₃	н	5-C1	OAc	0	MHe	
	4-OCF ₂ H	3-C1	5-C1	OMe	0	MMe	
35	4-C1	н .	5-c1	C(O)CH3	0	MMe	
	A-F	н	5-C1	C(0)NHMe	0	MHe	
	4-Br	H	5-C1	C(S)Me ₂	0	Mie	
40	4-CF ₃	н	5-F	Me	0	MMe	

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TABLE 5 (CONTINUED)

5	$\frac{R_1^2}{4-\text{OCF}_3}$	$\frac{R_1^b}{H}$	<u> </u>	<u>B</u> n-Bu	<u>×</u>	<u>E</u> Me	Phys. Prop. (m.p. °C)
	4-OCF ₂ H	3-C1	5-F	allyl	0	MHe	
10	4-C1 4-F	H H	5-F 5-F	i-Pr CH ₂ Ph	0	MMe MMe	
	4-Br 4-CF ₃	H H	5-F 5-CF ₃	Ph Ch ₂ ch ₂ F	0	nme Nme	
	4-OCF ₃	Н	5-CF ₃	CH ₂ OCH ₃	0	MMe	·
15	4-ocf ₂ H	3-C1	5-CF ₃	сн ₂ осн ₃	0	Me	
	4-C1	H	5-CF3	CH ₂ CN	0	Me	
	4-F	н	S-CF ₃	CH ₂ CO ₂ He	0	MMe	
20	4-Br	н	5-CF3	CS ₂ Me	0	Mie	
	4-CF ₃	Н	н	Не	S	Mile	
	4-0CF ₃	н	н	He	S	Mie	
25	4-OCF ₂ H	3-C1	H	He	S	Mie	
	4-CF ₃	Н	Н	He	0	MEt	
	4-CF ₃	н	5-F	CO ₂ Me	0	MEt	
	4-CF3	н	5-C1	Не	0	N-n-Bu	
30	4-CF ₃	н	н	CO ₂ Ne	0	¥-n-Bu	
	4-CF ₃	н	5-F	He	0	N-ally1	
	4-CF ₃	н	5-C1	CO ₂ Me	0	N-ally1	
J 5	4-CF ₃	н	н	Не	0	N-Ph	

TABLE 5 (CONTINUED)

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TABLE 6

	R ₁	R_	_			Phys.
5	4-CF ₃	$\frac{R_2}{H}$	<u>B</u> H	ţ O	Y Me	Prop. (m.p. *C)
	4-C1	H	Me	0	C(0)Me	
	4-0CF ₂ H	н	CO ₂ Me	0	n−Br	
10	4-CF ₃	н	C(O)He	0	SMe	
	4-C1	н	CO ₂ Et	0	SCC13	
	4-OCF ₃	Н	Н	1	Me	
	4-CF ₃	н	н	1	C(0)Me	oil
15	4-F	н	CO ₂ Me	1	Ме	
	4-OCF ₂ H	н	CO ₂ Me	1	C(0)Me	•
	4-CF ₃	Н	He	1	C(O)CF3	
20	4-C1	н	Ke	1	CO2Et	
	4-CF ₃	н	CO2CH2CF3	1	Me	
	4-OCF ₂ H	н	CO ₂ Et	1	Me	
25	4-0CF3	H	CO2CH2CH2CI	1	He .	
	4-Br	н	CONTINE	1	He	
	4-SCF ₂ H	Н	н	2	2-NO ₂ PhS	
	4-CF ₃	н	Ke	2	CO ₂ He	
30	4-C1	H	Me	2	C(O)Me	
	4-CF ₃	H	CO ₂ Me	2	C(0)Me	
	4-C1	Н	CO ₂ Me	2	He	
35	4-CF ₃	H	CO ₂ Me	1	C(0)Me	
		н	CO2He	1	He	glass
	4-CF ₃	н	CO ₂ Me	1	CO ₂ He	glass
	4-CF ₃	5-C1	CO2Re	1	C(O)Me	
40	4-CF ₃			1	Me	
	4-CF ₃	5-C1	CO ₂ Me	1	CO ₂ Me	
	4-CF ₃	4-F	CO ₂ He	1	C(O)Me	
45	4-CF ₃		CO2He	1	C(O)Me	
	4-CF ₃	4-F	CO ₂ Me	1	CO ₂ Me	
	4-CF ₃	5-F	CO ₂ He	1	CO ₂ He	
	4-CF3	н	н	1		178 to 179
50	3				•	

TABLE 7

5	Ð	D					Phys.
	<u>R</u> 1	<u>R</u> 2	B	<u>t</u>	¥	Ā	Prop. (m.p. °C)
	4-CF ₃	н	Не	0	He	0	
	4-CF3	Н	He	1	Нe	MMe	
10	4-CF ₃	5-C1	CO ₂ Me	2	Мe	8	
	4-CF ₃	5-C1	CO ₂ He	0	Мe	0	
		5-C1	Me	1	He	MHe	
15		5-C1	Me	2	C(O)Me	S	
	4-CF ₃	5-F	CO ₂ Me	0	C(O)He	0	
	_	5-F	CO ₂ Me	1	C(O)Me	MMe	
20	4-CF ₃	5-F	Не	2	C(O)He	so	
		5-F	Ме	0	C(O)Me	so ₂	
	4-CF3	H	CO ₂ Me	1	scc1 ₃	o .•	
	4-CF ₃	H	CO2He	2	SCC13	MMe	
25	4-CF ₃	н	Ме	0	CO2He	8	
	4-CF ₃	н	Не	1	CO2He	0	
	4-CF ₃	5-0CF ₂ H	CO ₂ He	2	n-Pr	Me	
30	4-CF ₃	5-OCF ₂ H	CO ₂ Me	0	n-Pr	S	•
	4-CF ₃	5-CF ₃	Же	1	C(0)CF ₃	0	
	4-CF ₃	5-CF3 .	Жe	2	C(0)CF3	MMe	•
35	4-CF ₃		CO ₂ Me	0	SMe	S	
			CO ₂ Me	1	SHe	0	
	4-CF ₃	H	н	1	Me	S	151 to 152
	4-CF ₃	H	н	1	CO2Me	S	150 to 158
40	4-CF ₃	H	н	1	C(O)He	S	80 to 85
		5-C1	н	1	C(0)He	S	97 to 99
	4-CF3	5-CL	н	1	CO ₂ Me	S	170 to 172
45	4-CF ₃	5-C1	н	1	Же	S	180 to 183
	4-C1	H	н	1	C(O)He	8	140 to 150
	4-C1	H	H	1	Xe	S	125 to 127
	4-Cl 4-Cl	5-C1 5-C1	H H	1	C(O)Me	S	127 to 142
50	7-04	2-01	п	•	He	•	160 to 165

TABLE 8

5	R ₁	R ₂	В	t	y	v	Phys. Prop. (m.p. °C)
J	R ₁ 4-CF ₃	H	<u>B</u> He	<u>t</u> 0	<u>у</u> Не	o O	
	4-CF ₃		Жe	1	Нe	WMe	
	4-CF3	5-C1	CO ₂ Me	2	Мe	S	
10	4-CF3	5-C1	CO ₂ Me	0	Нe	٥	
	4-CF3	5-C1	Нe	1	Нe	WMe	
	4-CF3	5-C1	Нe	2	C(O)Me	S	
15	4-CF3	5-F	CO ₂ Me	0	C(O)Me	0	
	4-CF3	5-F	CO ₂ He	1	C(0)Me	MMe	
	4-CF3	5-F	Нe	2	C(O)Me	SO	
20	4-CF3	5-F	He	0	C(O)He	so ₂	
20	4-CF3	н	CO ₂ He	1	scc1 ₃ .	٥	
	4-CF3	H	CO2Me	2	-	MMe	
	4-CF3	н	Жe	٥	CO ₂ Me		
25	4-CF3	н	Жe	1	CO_Me	0	
	4-CF ₃	5-OCF ₂ H	CO ₂ He	2	n-Pr	MMe	
	4-CF ₃	5-0CF_H	CO_He	0	n-Pr	S	
30	4-CF3	5-CF ₃	Me	1	C(O)CF ₃	0	
	4-CF ₃	5-CF ₃	Нe	2	C(0)CF ₃	MHe	
	4-CF ₃	5-C1	CO ₂ Me	0	SMe	S	
35	•	5-C1 .	CO2Me	1	SMe	0	

TABLE 9

	R.	R				Phys.
5	R ₁ 4-CF ₃	H ₂	<u>B</u> He	X He	<u>z</u> 0	Prop. (m.p. °C)
	4-C1		He	C(O)Me	MHe	
	4-CF ₃		Me	Et .	0	
10	4-C1	5-C1	Ke	CO ₂ Me	MHe	
	4-CF ₃	5-C1	Et	n-Pr	0	
	4-C1	5-F	Et	CO ₂ Et	MHe	
	4-CF ₃	5-F	Et	C(0)He	0	
15	4-C1	5-Cl	Et	s-cc1 ₃	MMe	
	4-CF ₃	5-01	(CH ₂) ₃ C1		0	
	4-C1			Ме	MMe	
žo	4-CF ₃		CH ₂ Ph	Me	0	
	4-C1	н	CH ₂ Ph		MMe	
	4-CF ₃	5-OCF ₃	Ph	C(O)CF ₃	0	
25	4-C1	5-OMe	Ph		MMe	
	4-CF ₃	5-C1	4-Cl-Ph		o	
	4-C1		4-C1-Ph	Et	MHe	
	4-CF ₃	Н	allyl	C(0)He	0	
30	4-C1	н	allyl	CO ₂ Et	MMe	
	4-CF ₃	5-C1	ОН	scc1 ₃	0	
	4-C1	н	OH	Хe	WMe	
	4-CF ₃	н	OAc	C(O)Me	0	
35	4-C1	н .	OAc .	He	WMe	

TABLE 10

	R	R				Phys.
5	R ₁ 4-CF ₃	H ₂	<u>B</u> Me	Y Ke	0	Prop. (m.p. °C)
	4-C1 4-CF ₃		Me Me	C(0)He Rt	MMe O	
10	_	5-C1	Ме	CO2He	MMe	
	4-CF ₃	5-C1	Et	n-Pr	0	
	4-C1	5-F	Et	COSEF	NMe	·
	4-CF ₃	5-F	Et	C(O)Me	0	
15	-	5-C1		S-CC13	MMe	
	4-CF ₃	5-C1	(CH ₂) ₃ C1	•	0	
	4-C1		(CH ₂)3C1		MMe	
20	4-CF ₃		CH ₂ Ph		0	
	4-C1		-	C(O)He	MMe	
	4-CF ₃	5-0CF ₃	Ph	C(O)CF ₃	.0	
25	_	5-0He	Ph	2-No, Phs	MHe	
	4-CF ₃	5-C1	4-C1-Ph	Жe	0	
			4-C1-Ph		MHe	
	4-CF ₃	н	allyl	C(0)He	0	
30	4-C1	н	allyl	CO2EF	WMe	
	4-CF ₃	5-C1	он	scc1 ₃	0	
	4-C1		OH	Me	NMe	
	4-CF ₃	н .	OAc	C(0)He	0	
35	4-C1	н	OAc	Ме	Y Me	

TABLE 11

5	R ₁ 4-CF ₃	$\frac{R_2}{H}$	<u>B</u> H	t O	<u>J</u> He	<u>K</u> H	Phys Prop		m.p. *C)
	4-C1	н	н		Re	 Н			
	4-CF ₃	Н	He	1 2	He	H			
10	4-C1	5-C1	Me	0	Ph	н.			
,,	4-CF ₃	5-01	CO ₂ Me	1	Ph	н			
	4-C1	5-C1	CO2Me	2	4-C1-Ph	H			
	4-CF ₃	H	н	0	4-F-Ph	н			
15	4-C1	н	н	1	4-CF ₃ -Ph	Н			
	4-CF ₃	н	Me	2	E t	н			
	4-C1	5-F	He	٥	n-Pr	н			
20	4-CF ₃	5-F	CO2He	1	n-Bu	H			
20	4-C1	5-F	CO ₂ Me	2	He	He			
	4-CF ₃	H	н	0	Не	Me			
	4-C1	H	н .	1	Me	Me			
25	4-CF ₃	H	He	2	He	Me			
	4-C1	5-C1	Нe	0	₽h	Me			
	4-CF ₃	5-C1	CO ₂ Me	1	Ph	Me			
	4-C1	5-C1	CO ₂ Me	2	Ph	He			
30	4-CF ₃	H	н	0	Ph	Me			
	4-C1	H	н	1	Et	Me			
	4-CF ₃	H	Нe	2	n-Pr	He			
	4-C1	H	Me	0	n-Bu	Me			
35	4-CF ₃	H	H	1	4-F-Ph	H	199	to	201
	4-CF3	H	Н	1	4-C1-Ph	н	187	to	188
	4-CF ₃	5-C1	H	1	4-F-Ph	н			
40	4-CF3	5-C1	H	1	4-C1-Ph	H			
	4-CF3	4-F	H	1	4-F-Ph	н			
	4-CF3	4-F	H	1	4-C1-Ph	H			
45		5-0CF2H	н	1	4-F-Ph	H			
. •	4-CP3	5-0CF2H		1	4-C1-Ph	H			
	4-C1	н	H	1	4-F-Ph	н	208	to	209
	4-C1	H	H	1	4-C1-Ph	H	179	to	181

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TABLE 12

	ъ	D						Phys.
5	$\frac{R_1}{4-CF_3}$	K2 H	<u>B</u> H	to	He 고	<u>K</u>	<u>v</u>	Prop. (m.p. °C)
			Нe	1	Нe	H	0	
			Et			н	٥	
10	4-CF ₃	H	CO,Me	0	Ph	Me	0	
	4-CF ₃	н	CO2Et	1	Мe	Me	0	
					He	Me	0	
15	4-CF ₃	4-C1	He	0	4-C1-Ph	He	0	
	4-CF3	Н	Et	1	He	H	S	
	4-CF ₃	H	CO ₂ He	2	He ·	H	S	
	4-CF3	4-Br	CO,Et	0	Ph	H	S	
20	4-CF ₃	н	н	1	Ph	He	S	
		н			Me	Me	S	
	4-CF ₃	H	Et	٥	He	He	S	
25	4-CF ₃	H	CO ₂ Me	1	4-F-Ph	He	S	
					Жe	H	MMe	
	4-CF ₃	Н	H	0	Жe	H	MMe	
30	4-CF ₃	5-C1	Ме	1	Ph	H	MMe	
	4-CF ₃	н	Zt	2	Ph	Мe	MHe	
	4-CF ₃	5-He	CO ₂ Me	0	Не	Мe	MHe	
35	4-CF ₃	н	CO ₂ Et	1	Me	Me	MMe	
3 0		н .	H		4-Cl-Ph	He	MMe	

TABLE 13

5	R ₁ .	H ₂	В	Ł	J	ĸ	v	Phys. Prop. (m.p. °C)
J	R ₁ A-CF ₃	ਸ	<u>B</u> H	<u>t</u> 0	<u>j</u> Me	<u>K</u> H	v O	110p. (m.pc)
		H	He	1	He	н	0	
	4-CF3	4-F	Et	2	Ph	н	0	
10	4-CF3	н	CO ₂ Me		Ph	He	0	
	4-CF3		COZEt		Жe	Нe	0	
	4-CF3	н	н	2	Me	He	0	
15	4-CF ₃	4-01	Нe	0	4-C1-Ph	He	0	
	4-CF ₃	н	Et	1	He	н	s	
	4-CF ₃	н	CO ₂ Me	2	He	H	S	
20	4-CF ₃		COZEŁ		Ph	H	s	
20	4-CF ₃	Н	H	1	Ph	Мe	S	
	4-CF ₃	н	He	2	He	Нe	S	
	4-CF ₃	H	Et	0	He	He	S	
25	4-CF ₃	н	CO ₂ Me	1	4-F-Ph	He	S	
	4-CF ₃		COZEŁ		Me	H	MMe	
	4-CF ₃	Н	H	0	He	H	WHe	
30	4-CF ₃	5-C1	Me	1	Ph	H	MMe	
			Et	2	Ph	Me	MMe	
	4-CF ₃	5-He	CO ₂ Me	0	Me	He	MMe	
35	_	H .	CO ₂ Et	1	He	Me	MMe	
٠	4-CF ₃	Н	Н	2	4-C1-Ph	He	MHe	

TABLE 14

5	R ₁ 4-CF ₃	$\frac{R_2}{H}$	B H	<u>J</u> Me	<u>к</u> н	<u>z</u>	Phys. Prop. (m.p. °C)
	4-C1 4-OCF H	H H	He He	He Ph	Ме Н	0	
10	4-CF ₃	н	allyl	4-C1-Ph	He	MMe	
	4-C1 4-OCF ₂ H	Н 5-F	allyl CH ₂ Ph	He 4-F-Ph	H Me	NMe NMe	
	4-CF3	H	CH ₂ Ph	Me	н	0	
15	4-C1 4-OCF ₂ H	H H	Ph Ph	Me Ph	Me H	0	
	4-CF3	Н	4-F-Ph	Ph	Me	Mile	
20	4-01 4-00F ₂ H	н 5-с1	4-P-Ph OH	He He	H Ke	NMe NMe	
	4-CF3	н	он	Нe	н	<u>,</u> 0	
	4-C1 4-OCF ₂ H	H H	OMe OMe	Et Ph	He H	0	
25	4-CF ₃	H	CO2Et	4-C1-Ph	Me	WHe	
	4-C1	H	COZEt	Me	H	Me	
	4-OCF ₂ H	5-C1	CONHMe	Ph	Нe	Mile	
30	4-CF ₃	Н	СОИНМе	Me	H	0	
	4-01 4-00F ₂ H	н н	n-Bu n-Bu	Ph Me	Me H	0	

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TABLE 15

5	R ₁ 4-CF ₃	H ₂	<u>B</u> He	<u>j</u> Ke	K K	<u>z</u>	Phys. Prop. (m.p. °C)
	4-C1	н	CO2Me	He	H	0	
	4-OCF ₂ H	Н	H_	Ke	H	0	
10	4-CF ₃	н	Ke	Ph	н	0	
	4-C1	н	CO2He	Ph	н	0	
	4-OCF ₂ H	Н	н	Ph	H	0	
15	4-CF ₃	5-C1	Et	4-C1-Ph	H	Me	
	4-C1	н	CO2Et	4-C1-Ph	H	MMe	
	4-COF ₂ H	н	(CH ₂) ₃ C1	4-C1-Ph	Н	Mie	
20	4-CF ₃	н	He	He	н	Me	
20	4-ocf ₂ h	H H	H CO ₂ He	He He	H Me	MMe O	
	4-CF ₃	н	н	Ke	He	.0	
25	4-C1 4-OCF ₂ H	H 5-F	He C(O)He	Me Ph	He He	0 .	
	4-CH ₃	н	C(O)Et	Ph	Жe	0	
30	4-C1 4-OCF ₂ H	H H	He Et	Ke Ke	He He	NMe Mie	
	4-CF ₃	н	n-Pr	Ph	Me	Me	
	4-C1	н	CO2He	Ph	He	Mie	
35	4-0CF ₂ H	н	_	4-C1-Ph	Me	MMe	
J.J	4-CF3	H	Et	4-C1-Ph	Me	MHe	

TABLE 16

- 5	ï
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	R	R ₂	<u>B</u>	R _B	R _B	R.c	R ^d 8	Phys. Prop. (m.p. °C)
10	4-CF ₃	Н	H	. He	H	F.	н	
	4-C1 .	H	н	He	н	н	н	
	4-Br	H	H	Me	H	H	H	
	4-ocf ₂ h	H	Н	He	н	H	К	
15	4-CF ₃	н	CO ₂ Me	Жe	H	14	Н	
	4-C1	Н	COZHe	Me	H	h	H	
	4-Br	Н	CO ₂ He	Me	Н	н	Н	
20	4-OCF ₂ H	н	CO ₂ Me	Me	H	Н	н	
	4-CF ₃	5-C1	н	He	н	н	Н	
	4-C1	5-C1	H	He	H	Н	Н	
	4-Br	5-C1	H	Мe	H	H	Ĥ	
25	4-OCF ₂ H	5-C1	Н	Me	H	H	Н	
	4-CF ₃	5-C1	CO ₂ Me	Me	Н	H	Н	
	4-C1	5-C1	CO2He	Нe	н	H	н	
30	4-Br	5-C1	CO2He	Мe	н	F "	Н	
	4-OCF ₂ H	5-C1	CO ₂ Me	Me	н	H	H	
	4-CF ₃	4-F	н	He	н	H	н	
	4-C1	4-F	Н	Мe	Н	::	Н	
35	4-Br	4-F	Н	Жe	н	h	Н	
	4-OCF ₂ H	4-F	Н	He	H	11	н	
	4-CF ₃	4-F	CO ₂ Me	He	H	H	Н	
	4-C1	A-F	CO2He	Нe	H	H	H	
40	4-Br	4-F	CO ₂ Me	Мe	H	H	H	
	4-OCF ₂ H	4-F	CO ₂ Me	Нe	H	H	Н	
	4-CF ₃	н	н	Нe	Мe	H	н	
45	4-C1	H	H	Жe	Мe	H	Н	
	4-Br	H	H	Мe	Мe	H	н	
	4-0CF ₃	Н	Н	He	He	H	Н	
	4-CF ₃	H	CO ₂ Me	Жe		H		
50	4-C1	н	CO2Me	Бė	He	H	H	
	4-Br	H	CO,Me	Нe	Жe	H	H	
	4-OCF ₃	н	CO ₂ Me	Me	Нe	H	H	
55	4-CF ₃	н 5-ост ₂ н	H	He	Мe	Н	н	
	4-C1	5-OCF ₂ H	н	Нe	Мe	H	H	

TABLE 16 (continued)

	R.	R_	_	R [®]	R ₈	R ₈	R d 8	Phys.
	<u>R</u> 1	<u>R</u> 2	B	R ₈	8	<u>~8</u>	<u>~8</u>	Prop. (m.p. °C)
5	4-Br	5-0сғ ₂ н	H	Me	Ke	H	H	
	4-OCF ₃	5-0CF ₂ H	H	He	Me	H	H	
	4-CF ₃	5-0CF ₂ H	CO2Ne	Ke	He	H	н	
	4-C1	5-0CF ₂ H	CO2He	Мe	Me	H	н	
10	4-Br	5-0cf ₂ h	CO2He	Мe	Me	н	н .	
	4-0CF ₃	5-00F ₂ H	CO ₂ Me	Жe	Ke	н	H	
	4-CF ₃	5-C1 ²	H 2	Me	Ke	н	Н	
15	4-C1	5-C1	н	Me	Мe	н	H	
	4-Br	5-C1	н	He	Ke	H	н	
	4-0CF3	5-C1	н	He	Нe	H	H	
	4-CF ₃	5-C1	CO ₂ He	Нe	Ke	н	H	
20	4-C1	5-C1	CO2He	Мe	He	н	н	
	4-Br	5-C1	CO2He	Мe	Ke	н	н	
	4-0CF ₃	5-C1	CO2He	Мe	Me	н	H	
25	4-CF ₃	н	н	Н	He	н	H	
	4-C1	н	H	н	Жe	н	н	
	4-CF ₃	Н	CO ₂ Me	H	He	H	H	
	4-C1	н	CO ₂ Me	н	He	Н	H	
30	4-CF ₃	5-C1	H	н	He	н	н	
	4-C1	5-C1	H	н	Me	н	H	
	4-CF ₃	5-C1	CO Me	H	Жe	н	H	
	4-C1	5-C1	CO2He	н	Нe	н	н	
35	4-CF ₃	H	н	н	H	He	н	
	4-C1	н	H	H	H	He	н	
	4-CF ₃	H	CO2He	н	H	He	н	
40	4-C1	H	CO2Ne	н	н	Жe	н	
	4-CF ₃	5-C1	_	H	н	Me	н	
	4-C1	5-C1	н	н	н	He	H	
	4-C1 4-CF ₃	5-C1	H CO ₂ Ne	н	H	Хe	H	
45	4-C1	5-C1	CO, Me	н	н	Мe		
	4-CF ₃	H	н	H	н	н		
	4-C1	н				н	He	
	4-C1 4-CF ₃	H	H CO ₂ He	H	H		Me	
50	4-C1	H	CO ₂ He	н	н	н	He	
	4-CP ₃	5-C1	н			H	He	
	4-C1	5-C1	н	H	н	н	Me	
55	4-CF ₃	5-C1	H CO ₂ He	H	H	H	He	
	4-CF ₃ 4-C1	5-C1	CO2He	н	н	н	He	
			-					

			IYR	LE 16	(हरूत	timued)		
5	R	<u> </u>	B	R 8	R ₈	R ₈	Rd B	Phys. Prop. (m.p. *C)
	CF ₃	н	H	H	н	He	Me	
	Cl	н	н	H	н	Me	Me	
	Br	H	H CO ₂ Me	H	H	Не	He	
10	CF ₃	н	202	H	H	Me	Ke	
	Cl	Н	CO ₂ Me		H	Мe	Xe	
	Br		CO ₂ Me		H	Me	He	
15	CF ₃	5-C1	Н	н	H	He	Мe	
		5-C1	H	H	H	He	Ke	
	Br CF ₃	5-C1 5-C1	H CO ₂ Me	H	H H	Me Me	Xe Xe	
	C1		CO ₂ Me		н	Me	Нe	
20	Br		CO2He			Жe	Me	
		5-OCF ₂ H	_	н	н	Me	Me	
		5-0CF ₂ H		H	н	Xe	Me	
25		5-0CF ₂ H				Me	Me	
		5-0CF ₂ H			н	Me	He	
		5-0CF ₂ H	_			He .	He	
30			CO ₂ Me		н	Me	He	
30	CF ₃	4-F	H	н	н	He	Жe	
	C1	4-F	н	н	H	He	Me	
	Br	4-F	H	H	H	Ме	Жe	
35	CF ₃	4-F	CO ₂ Me	H	H	Мe	Me	
	Cl		CO ₂ He		H	He	Me	
	Br		CO2He		н	He	Me	
40	4-CF ₃		CO2He		н	H	H	
~•	4-CF3	H	CO2Me	H	i-Pr	H	H	
	4-CF3	н	CO2Me	H	н	Ph	H	
	4-CF ₃	н	CO2Me	H	H	н	Ph	
45	4-CF ₃	5-C1	CO2Me				н	
	4-CF3	5-C1	CO2Me	H	Cl	H	H	
	4-cg		CO2He				н	
50	A-CF ₂	5-C1	COMe	H	P	н	H	
	4-CF3	5-F	CO_He	H	н		H	
				н	н	н	CO2He	
	<u> </u>		4				4	

TABLE 16 (continued)

	R	<u>R</u> 2	B	R ₈	R _B	R ₈	R ₈	Prop. (m.p. °C)
5	4-CF ₃	5-7	CO, Me	Et	H	н	н	
	4-CF ₃	5-7	CO2Me	н	Et	H	н	
	4-CF ₃	5-0CF ₂ H	CO ₂ Me	Ph	Н	н	H	
10	4-CF ₃	5-0CF2H	CO ₂ He	H	Ph	H	H	
	4-CF ₃	5-OCF ₂ H	CO ₂ Me	H	Ke	Мe	H	
	4-CF ₃	5-0CF2H	CO ₂ Me	He	н	н	He	
	4-CF ₃	4-C1	CO ₂ Me	i-Pr	H	H	H	
15	4-CF ₃	4-C1	CO ₂ Me	H	i-Pr	Н	H	
	4-CF ₃	4-C1	CO2He	Et	Et	Н	н	
	4-CF ₃	4-01	CO ₂ He	Et	He	H	H	
20	4-CF ₃	4-F	CO2Ne	Br	H	H	H	
	4-CF ₃	4-F	CO ₂ Me	H	Br	H	H	
	4-CF ₃	4-F	CO ₂ Me	i-Pr	H	H	H	
25	4-CF ₃	4-F	CO2Ne	H	i-Pr	H .	H .	

JQ

35

40

45

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			TAI	BLE 1	<u> </u>		•
5	R	R ₂	B	¥	R 8	R _B	Phys. Prop. (m.p. °C)
	4-CF ₃	н	н	0	Нe	н	
	4-C1	н	H	0	Мe	H	
10	4-Br	H	H	0	He	H	
	4-ocf ₂ H	н	H	0	Нe	H	
	4-CF ₃	н	H	0	Н	Нe	
	4-C1	Н	н	0	H	Нe	
15	4-Br	н	н	0	н	Нe	
	4-OCF ₂ H	Н	н.	0	Н	Me	
	4-CF ₃	Н	Н	0	Жe	Жe	
	4-C1	н	H	0	Me	Жe	
20	4-Br	H	H	0	Жe	Мe	
	4-OCF ₂ H	Н	н	0	Ме	He	
	4-CF ₃	Н	Н	0	Ph	Н	
	4-C1	Н	H	0	Ph	н	
25	4-Br	Н	н	0	Ph	H	
	4-OCF ₂ H	Н	Н	0	Ph	н -	
	4-CF ₃	н	н	0	н	Ph	
	4-C1	H	H	0	Н	Ph	
30	4-Br	Н	H	0	н	Ph	
	4-OCF ₂ H	н	Н	0	Н	Ph	
	4-CF ₃	н	н	0	CO2Ne	H	
25	4-C1	Н	Н	0	CO2Ne	H	
35	4-Br	н	Н	0	H	CO ₂ Me	
	4-OCF ₂ H	н .	H	0	H	CO ₂ He	
	4-CF ₃	5-C1	Н	0	Мe	Н	
40	4-C1	5-C1	Н	0	Ke	н	
	4-Br	5-C1	н	0	Вe	Н	
	4-OCF ₂ H	5-C1	Н	C	Ke	Н	
	4-CF ₃	5-C1	Н	0	Н	Ме	
45	4-C1	5-Cl	Н	0	Н	Me	
	4-Br	5-C1	Н	0	H	Ke	
	4-OCF ₂ H	5-C1	Н	0	н	Me	
	4-CF ₃	5-C1	Н	0	Жe	Мe	
50	4-C1	5-C1	н	0	Жe	Мe	
	4-Br	5-C1	н	0	Жe	Жe	
	4-OCF ₂ H	5-C1	Н	Ο.	Жe	Me	
	4-CF ₃	5-C1	н	0	Ph	н	
55	4-C1	5-C1	Н	0	Ph	H	
	4-Br	,5-C1	н	0	Ph	Н	

		TABLE 17 (continued)									
5	<u>R</u> 1	R ₂	B	<u>v</u>	R _B	R ₈	Phys. (m.p. *c)				
	4-OCF ₂ H	5-C1	н	0	Ph	н					
	4-CF3	5-C1	н	0	Н	Ph					
10	4-C1	5-C1	н	0	н						
,0	4-Br	5-C1	 Н	Ö	Н	PH Ph					
	4-OCF ₂ H	5-C1	H	0	• н	Ph					
	4-GF3	5-C1	н	0	CO ₂ Me	н					
15	4-C1	5-C1	н	0	CO ₂ Me	Н					
	4-Br	5-C1	н	0	н	CO2He					
	4-OCF ₂ H	5-C1	Н	0	H	CO ₂ Me					
20	4-CF ₃	Н	н	s	Me	н	208 to 210				
	4-61	н	н	S	Мe	н	172 to 178				
	4-Br	Н	H	S	Me	H	185 to 187				
	4-OCF ₂ H	Н	н	S	Me	H					
25	4-CF3	H	н	S	H	He .	168 to 170				
	4-C1	н	H	S	H	Me	155 to 156				
	4-Br	Н	H	S	н	Me	172 to 180				
	4-OCF ₂ H	Н	н	S	н	He					
30	4-CF ₃	н	н	S	Me	Нe	183 to 184				
	4-C1	H	Н	S	Me	Жe	181 to 186				
	4-Br	H	H	S	He	Жe	182 to 185				
	4-OCF ₂ H	Н	H	S	Me	He					
35	4-CF ₃	н	н	S	Ph	H					
	4-C1	Н	H	S	Ph	H					
	4-Br	н	H	S	Ph	н					
	4-OCF ₂ H	н	н	S	Ph	H					
40	4-CF ₃	Н	н	S	Н	Ph					
	4-01	н	Н	S	H	Ph					
	4-Br	н	H	S	H	Ph					
	4-OCF ₂ H	н	н	S	Н	Ph					
45	4-CF ₃	н	H		-	Н					
	4-C1	Н	н	\$	CO2He	Н					
	4-Br	H .	н	S	н	CO2He					
50	4-OCF ₂ H	Н	н	S	н	CO2He					
	4-CF ₃	5-C1	н	S	He	H	251 to 255				
	4-C1	5-C1	H	S	Me	H	222 to 226				
	4-Br	5-C1	Н	s	Нe	H	228 to 230				
55	4-OCF ₂ H	5-C1	Н	S	Мe	H					
) 5	4-CF ₃	5-C1	н	S	н	Me	171 to 173				

5	TABLE 17 (continued)								
	<u>R</u> 1	<u>R</u> 2	<u>B</u>	<u>v</u>	R. 8	R ₈	Phys Prop		m.p. *C)
	4-C1	5-C1	H	8	H	He	170		
10	4-Br	5-C1	H	S	Н	He	120	to	125
	4-OCF ₂ H	5-C1	H	\$	H	He			
	4-CF ₃	5-C1	H	8	He	He	200		
	4-C1	5-C1	H	S	Не	Хe	190		
15	4-Br	5-C1	H	5	He	Нe	204	to	207
	4-OCF ₂ H	5-C1	H	8	Me	Мe			
	4-CF ₃	5-C1	H	S	Ph	H			
20	4-C1	5-C1	H	S	Ph	н			
20	4-Br	5-C1	H	S	Ph	H			
	4-OCF ₂ H	5-C1	Н	S	Ph	Н			
	4-CF ₃	5-C1	H	S	H	PH			
25	4-C1	5-C1	H	S	H	Ph			
25	4-Br	5-C1	H	S S	Н	Ph			
	4-OCF ₂ H	5-C1	H		H	Ph			
	4-CF ₃	5-C1	H	S	CO ₂ He	н -			
30	4-C1	5-C1	Н	8	Co ₂ Me	н			
	4-Br	5-C1	Н	S	Н	CO ₂ Me			
	4-OCF ₂ H	5-C1	Н	S	н	CO ₂ Ke			
	4-CF ₃	5-Br	H	S	Не	н			
35	4-C1	5-Br	H	S	Нe	H			
	4-Br	5-Br	H	S	Мe	H			
	4-CF ₃	5-F	H	S	Me	H			
	4-C1	5-F.	H	S	Же	H			
40	4-Br	5-F	H	2 2	Xe	H			
	4-CF ₃	4-F	H		Xe	H			
	4-C1	4-7	H	8	Жe	H			
	4-Br	4-P	H	8	Не	H			
45	4-CF ₃	5-Br	H	8	H	Ke			
	4-C1	5-Br	H	8	Н	Нe			
-	4-Br	5-Br	H	S	H	He Ma			
	4-CF ₃	5-7	Н	\$	Н	He			
50	4-C1	5-F	H	8	H	Xe			
50	4-Br	5-F	H	8	H	He			
	4-CF ₃	4-F	н	S	H	Me			
	4-C1	4-F	H	8	H	Ke Ke			
55	4-Br 4-CF ₃	4-F 5-Br	H H	2 2	H Me	Же Ķе	190	to	196
- -	3 4-Cl	5-Br	н	S	Ke	Жe	165		

			TAE	SLE 17	(contin	ued)	
5	R ₁	R ₂	B	¥	R 8	R ₈	Phys. Prop. (m.p. °C)
	4-Br 4-CF ₃	5-Br 5-F	H H	S S	Me Me	He He	208 to 211
10	4-Cl 4-Br	5-F 5-F 4-F	н н н	s s	He He He	He He He	
	4-CF ₃ 4-C1 4-Br	4-F 4-F	н н	s s	Me Me	Me Me	
15	4-CF ₃	4-C1	Н	S	He	Мe	170 to 173
	4-Cl 4-Br	4-C1 4-C1	H H	s S	Me Me	He He	150 to 168 158 to 170
20				TA	BLE 18		
	<u>R</u> 1	R ₂	B	<u>v</u>	R ₈	R _p	Phys. Prop. (m.p. °C)
25	4-CF ₃	н	н	0	Me	н	
	3		CO2Ne	0	He	н	
	3	5-C1	Н	\$	Мe	H	
30	4-CF ₃	5-C1	CO2Me	S	н	Me	
	4-CF ₃		Н	0	Н	He	
	4-CF ₃		CO2He	0	Н	Me	
35	4-CF ₃		н	S	Нe	Нe	
	4-CF ₃	5-F	CO ₂ Me	S	Нe	Me	
		5-0CF ₂ H		0	Нe	Me	
40		5-0CF ₂ H				He	
	3	4-01	H	S	H	He	
	4-CF ₃ 4-Cl		CO ₂ Me	S	н	He	
45	4-C1	H H	H CO ₂ He	0	Me Me	He Me	
	4-C1 4-C1	5-Cl 5-Cl	H CO Me	S	He	Me	
	4-C1	5-C1	H H	S	н н	He He	
	4-C1	5-C1	CO2Me	0	н	He	
50	4-C1	5-ocf ₂ H	Н	S	Мe	He	
	4-C1		CO ₂ Me	\$	Нe	Жe	
	4-C1	5-00F ₂ H	H	0	Нe	Me	
55	4-C1	5-OCF ₂ H	CO2He	0	н	Me	
	4-C1 4-C1	4-F 4-F	H CO ₂ He	s s	H H	He He	

Formulation and Use

The compounds of this invention will generally be used in formulation with an agriculturally suitable carrier comprising a liquid or solid diluent or an organic solvent. Useful formulations of the compounds of Formula I can be prepared in conventional ways. They include dusts, granules, pellets, solutions, suspensions, emulsions, wettable powders, emulsifiable concentrates, dry flowables and the like. Many of these can be applied directly. Sprayable formulations can be extended in suitable media and used at spray volumes of from about one to several hundred liters per hectare. High strength compositions are primarily used as intermediates for further formulation. The formulations, broadly, contain about 1% to 99% by weight of active ingredient(s) and at least one of a) about 0.1% to 20% surfactant(s) and b) about 5% to 99% solid or liquid diluent(s). More specifically, they will contain effective amounts of these ingredients in the following approximate proportions:

15		Active	Percent by Weight			
			<pre>Diluent(s)</pre>	Surfactant(s)		
20	Wettable Powders	25-90	0-74	1-10		
	Oil Suspensions, Emulsions, Solutions, (including Emulsifian Concentrates)		40-95	0-15		
25	Dusts	1-25	70-99	0-5		
	Granules and Pellets	1-95	5-99	0-15		
30	High Strength Compositions	90-99	0-10	0-2		

Lower or higher levels of active ingredient can, of course, be present depending on the intended use and the physical properties of the compound. Higher ratios of surfactant to active ingredient are sometimes desirable, and are achieved by incorporation into the formulation or by tank mixing.

Typical solid diluents are described in Watkins, et al., "Handbook of Insecticide Dust Diluents and Carriers", 2nd Ed., Dorland Books, Caldwell, New Jersey. The more absorptive diluents are preferred for wettable powders and the denser ones for dusts. Typical liquid diluents and solvents are described in Marsden, "Solvents Guide," 2nd Ed., Interscience, New York, 1950. Solubility under 0.1% is preferred for suspension concentrates; solution concentrates are preferably stable against phase separation at 0°C. "McCutcheon's Detergents and Emulsifiers Annual", Allured Publ. Corp., Ridgewood, New Jersey, as well as Sisely and Wood, "Encyclopedia of Surface Active Agents", Chemical Publ. Co., Inc., New York, 1964, list surfactants and recommended uses. All formulations can contain minor amounts of additives to reduce foam, caking, corrosion, microbiological growth, etc. Preferably, ingredients should be approved by the U.S. Environmental Protection Agency for the use intended.

The methods of making such compositions are well known. Solutions are prepared by simply mixing the ingredients. Fine solid compositions are made by blending and, usually, grinding as in a hammer or fluid energy mill. Suspensions are prepared by wet milling (see, for example, Littler, U.S. Patent 3.060,084). Granules and pellets may be made by spraying the active material upon preformed granular carriers or by agglomeration techniques. See J. E. Browning, "Agglomeration". Chemical Engineering, December 4, 1967, pages 147 and following, and "Perry's Chemical Engineer's Handbook", 4th Ed., McGraw-Hill, New York, 1963, pages 8 to 59 and following.

Examples of useful formulations of compounds of the present invention are as follows:

Example A

Emulsifiable Concentrate

3.3a,4,5-tetrahydro-3a-methyi-N-[4-(trifluoromethyi)phenyl]-2H-benz[g]indazole-2-carboxamide 5% blend of oil soluble sulfonates and polyoxyethylene ethers 4% xylene 91%

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The ingredients are combined and stirred with gentle warming to speed solution. A fine screen filter is included in packaging operation to insure the absence of any extraneous undissolved material in the product.

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Example B

15 Wettable Powder

Methyl 3.3a,4,5-tetrahydro-2-[[4-(trifluoromethyl)phenylamino]carbonyl]-2H-benz[g]indazole-3a-carboxylate 30%

sodium alkylnaphthalenesulfonate 2% sodium ligninsulfonate 2% synthetic amorphous silica 3% kaolinite 63%

The active ingredient, warmed to reduce viscosity, is sprayed onto the inert materials in a blender. After grinding in a hammer-mill, the material is reblended and sifted through a 50 mesh screen.

Example C

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Dust

Wettable powder of Example 8 10% pyrophyllite (powder) 90%

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The wettable powder and the pyrophyllite diluent are thoroughly blended and then packaged. The product is suitable for use as a dust.

40 Example D

Granule

3.3a.4,5-tetrahydro-3a-methyl-N-[4-(trifluoromethyl)phenyl]-2H-benz[g]indazole-2-carboxamide 10% attapulgite granules (low volative matter, 0.71-0.30 mm; U.S.S. No. 25-50 sieves) 90%

The active ingredient is dissolved in a suitable solvent and sprayed onto dedusted attapulgite granules in a double cone blender. The granules are warmed to drive off solvent, cooled and packaged.

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Example E

55 Granule

Wettable powder of Example D 15% gypsum 69% potassium sulfate 16%

The ingredients are blended in a rotating mixer and water sprayed on to accomplish granulation. When most of the material has reached the desired range of 0.1 to 0.42 mm (U.S.S. No. 18 to 40 sieves), the granules are removed, dried, and screened. Oversize material is crushed to produce additional material in the desired range. These granules contain 4.5% active ingredient.

Example F

10

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Solution

Methyl 3,3a,4,5-tetrahydro-2-[[4-(trifluoromethyl)phenylamino]carbonyl]-2H-benz[g]indazole-3a-carboxylate 10%

isophorone 90%

The ingredients are combined and stirred to produce a solution suitable for direct, low volume application.

Compounds of Formula I can also be mixed with one or more other insecticides, fungicides, nematocides, bactericides, acaricides, or other biologically active compounds to form a multi-component pesticide giving an even broader spectrum of effective agricultural protection. Examples of other agricultural protectants with which compounds of the present invention can be mixed or formulated are:

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Insecticides:

3-hydroxy-N-methylcrotonamide(dimethylphosphate)ester (monocrotophos) methylcarbamic acid, ester with 2,3-dihydro-2,2-dimethyl-7-benzofuranol (carbofuran)

O-[2.4,5-trichloro-α-(chloromethyl)benzyl]phosphoric acid, O',O'-dimethyl ester (tetrachlorvinphos)
2-mercaptosuccinic acid, diethyl ester, S-ester with thionophosphoric acid, dimethyl ester (malathion)
phosphorothicic acid, O,O-dimethyl, O-p-nitrophenyl ester (methyl parathion)
methylcarbamic acid, ester with α-naphthol (carbaryl)
methyl O-(methylcarbamoyl)thiolacetohydroxamate (methomyl)

N'-(4-chloro-o-tolyl)-N,N-dimethylformamidine (chlordimeform)
O,O-diethyl-O-(2-isopropyl-4-methyl-6-pyrimidylphosphorothioate (diazinon)

octachlorocamphene (toxaphene)

O-ethyl O-p-nitrophenyl phenylphosphonothioate (EPN)

(S)- α -cyano- $\underline{\mathbf{m}}$ -phenoxybenzyl(1R,3R)-3-(2,2-dibromovinyl)-2,2-dimethylcyclopropanecarboxylate (deltamethrin)

Methyl N',N'-dimethyl-N-[(methylcarbamoyl)oxy]-1-thiooxamimidate (oxamyl) cyano(3-phenoxyphenyl)-methyl-4-chloro- α -(1-methylethyl)benzeneacetate (fenvalerate) (3-phenoxyphenyl)methyl(\pm)- α -(2,2-dichloroethenyl)-2.2-dimethylcyclopropanecarboxylate (permethrin)

α-cyano-3-phenoxybenzyl 3-(2.2-dichlorovinyl)-2,2-dimethylcyclopropane carboxylate (cypermethrin)
 O-ethyl-S-(p-chlorophenyl)ethylphosphonodithioate (profenofos)
 phosphorothiolothionic acid, O-ethyl-O-[4-(methylthio)phenyl]-S-n-propyl ester (sulprofos).
 Additional insecticides are listed hereafter by their common names: triflumuron, diflubenzuron, methoprene, buprofezin, thiodicarb, acephate, azinphos-methyl, chlorpyrifos, dimethoate, fonophos, isofenphos, methidathion, methamidiphos, monocrotophos, phosmet, phosphamidon, phosalone, pirimicarb, phorate, profenofos, terbufos, trichlorfon, methoxychlor, bifenthrin, biphenate, cyfluthrin, fenpropathrin, fluvalinate, flucythrinate, tralomethrin, metaldehyde and rotenone.

55 Fungicides:

methyl 2-benzímidazolecarbamate (carbendazim) tetramethylthiuram disulfide (thiuram) n-dodecylguanidine acetate (dodine)

manganese ethylenebisdithiocarbamate (maneb) 1.4-dichloro-2,5-dimethoxybenzene (chloroneb) methyi 1-(butylcarbamoyl)-2-benzimidazolecarbamate (benomyl) 1-[2-(2.4-dichlorophenyl)-4-propyl-1.3-dioxolan-2-ylmethyl]-1H-1.2.4-triazole (propiconazole) 2-cyano-N-ethylcarbamoyl-2-methoxyiminoacetamide (cymoxanil) 1-(4-chlorophenoxy)-3,3-dimethyi-1-(1H-1,2,4-triazol-1-yl)-2-butanone (triadimefon) N-(trichloromethylthio)tetrahydrophthalimide (captan) N-(trichloromethylthio)phthalimide (folpet) 1-[[[bis(4-fluorophenyl)][methyl]silyl]methyl]-1H-1,2,4-triazote. 10

Nematocides:

S-methyl 1-(dimethylcarbamoyl)-N-(methylcarbamoyloxy)thioformimidate S-methyl 1-carbamoyl-N-(methylcarbamoyloxy)thioformimidate

N-isopropylphosphoramidic acid, O-ethyl O'-[4-(methylthio)-m-tolyl]diester (fenamiphos).

Bactericides:

tribasic copper sulfate streptomycin sulfate.

Acaricides:

senecioic acid, ester with 2-sec-butyl-4,6-dinitrophenol (binapacryl)

25 6-methyl-1.3-dithiolo[4.5-ß]quinoxalin-2-one (oxythioquinox)

ethyl 4.4'-dichlorobenzilate (chlorobenzilate)

1.1-bis(p-chlorophenyl)-2.2,2-trichloroethanol (dicofol)

bis(pentachloro-2.4-cyclopentadien-1-yl) (dienochlor)

tricyclohexyltin hydroxide (cyhexatin)

trans-5-(4-chlorophenyl)-N-cyclohexyl-4-methyl-2-oxothiazolidine-3-carboxamide (hexythiazox)

amitraz

propargite

fenbutatin-oxide

bisclofentezin.

35

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Biological

Bacıllus thuringiensis Avermectin B.

Utility

The compounds of the present invention exhibit activity against a wide spectrum of foliar and soil 45 inhabiting insects. Those skilled in the art will recognize that not all compounds are equally effective against all insects, but compounds of this invention display control of many of the economically important pest species of the insect orders Lepidoptera, Homoptera, and Coleoptera among many others. The specific species for which control is exemplified below are: fall armyworm, Spodoptera frugiperda: boll weevil, Anthonomus grandis; European corn borer, Ostrinia nubilalis; southern corn rootworm, Diabrotica undecimpunctata howardi; aster leafhopper. Macrosteles fascifrons; Tobacco budworm, Heliothis virescens. The pest control afforded by the compounds of the present invention is not limited, however, to these species.

Application

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Insects are controlled and agricultural drops are protected by applying one or more of the Formula I compounds of this invention, in an effective amount, to the locus of infestation, to the area to be protected,

or directly on the pests to be controlled. A preferred method of application is by spraying with spray equipment that distributes the compound on the foliage, in the soil, or to the plant part that is infested or needs to be protected. Alternatively, granular formulations of these compounds can be applied to soil or foliage or, optionally, incorporated into the soil. Either aerial or ground application can be used.

The substituted indazole(s) of this invention can be applied in its(their) pure state, but most often application will be of a formulation comprising one or more compounds of this invention, in an agriculturally suitable carrier or diluent. A most preferred method of application involves spraying a water dispersion or refined oil solution of the compounds.

The rate of application of the Formula I compounds required for effective control will depend on such factors as the species of insect to be controlled, the pest's life stage, its size, its location, the nost crop, time of year of application, ambient moisture, temperature conditions, and the like. In general, application rates of 0.05 to 2 kg of active ingredient per hectare are sufficient to provide effective control in large scale field operations under normal circumstances, but as little as 0.01 kg/hectare may be sufficient or as must as 8 kg/hectare may be required, depending upon the factors listed above.

The following Examples demonstrate the control efficacy of compounds of Formula I on specific insect pests wherein Compounds 1 through 8 are those of Examples 1 to 8 and Compounds 9 through 181 are as depicted in Tables 19 and 20. Compounds not listed were either not tested or did not effect at least 80% mortality in that screen.

TABLE 19

5			4				
		R		•			
		2 .((~J			
10			, <u>"</u> "–"	!	R.		
			Ĩ	2	* 1 · 1		
			00	_n-1) > *		
15				Y	•		
	CMPD	R	<u>R</u> 2	ī	Y	<u>t</u>	B
	9	4-C1	н	н	н	1	н
20	10	4-Br	н	H	H	1	H
	11	4-OCF ₃	н	н	н	1	H
	12	4-C1	5-C1	н	н	1	H
	13	4-Br	5-C1	H	H .	1	H
25	14	4-CF ₃	5-OHe	H	H		
	15	4-CF ₃	5-ocf ₂ h	Н	H	1	Н
	16	4-CF ₃	5- F	Н	H	1	н
30	17	4-CF ₃	4-C1	н	H	1	Н
,,	18	4-CF ₃	4-F	H	H	1	Н
	19	4-CF ₃	5,6-di-Cl	Н	н	1	н
	20	4-CF ₃	5- F	Н	н	1	CO ₂ He
35	21	4-CF ₃	5-0Me	H	н	1	CO2He
	22	4-CF ₃	4-C1	H	H .	1	CO2He
	23	4-CF ₃	4-F	H	H	1	CO ₂ Me
40	24	4-CF ₃	5,6-di-C1	H	H	1	CO2He
	25	4-CF ₃	Н	H	H	1	Et
	26	4-CF ₃	н	H	Н	1	CH ₂ -Ph-4-F
4 5	27	4-CF ₃	н	H	Н	1	(CH ₂) ₂ CN
-5	28	4-CF ₃	Н	H	Н	1	(CH ₂)3CM
	29	4-CF ₃	Н	H	н	1	CO2Et
	30	4-CF ₃	H	H	н	1	CO2(CH2)2C1
50	31	4-CF	н	H	н	1	co,(CH,),Br

TABLE 19 (continued)

	CMPD	R	<u>R</u> 2	<u> </u>	Y	<u> </u>	<u>B</u>
	32	4CF ₃	н	н	н	1	CO2CH2CF3
5	33	4-CF ₃	н	н	H	1	co2cH2cc13
	34	4-CF ₃	H	H	н	1	CONH
	35	4-CF ₃	н	н	н	1	CONHMe
10	36	4-CF ₃	Н	н	н	1	COMMe ₂
	37	4-CF ₃	H	н	н	1	CON(H)CH2CH=CH2
	38	4-CF ₃	н	Н	н	1	CON (Me) CH2 CH= CH2
15	39	4-CF ₃	н	н	н	1	CONH(4-C1-Ph)
	40	4-C1	5-C1	H	н	1	co ₂ Me
	41	4-Br	5-C1	Н	H	1	CO ₂ Me
20	42	4-CF ₃	H	Н	н	1	CO-piperidinyl
20	43	4-CF ₃	H	H	H	1	∞ ₂ н
	44	4-CF ₃	Н	н	H	1	CO ₂ Na
	45	4-CF ₃	н	н	C(0)Me	1	CO ₂ Me
25	46	4-CF3	H	H	Me	1	CO ₂ Me
	47	4-CF ₃	H	н	CO ₂ Me	1	CO ₂ Me
	48	4-CF ₃	H	H	C(O)Me	1	н
30	49	4-CF ₃	H	4-F-Ph	н	1	н
	50	4-C1	Н	4-F-Ph	Н	1	H
	51	4-CF ₃	H	4-C1-Ph	H	1	н
	52	4-C1	H	4-C1-Ph	H	1	H
35	53	4-CF ₃	H	Н	Н	1	CH2CH=CH2
	54	4-CF3	H .	H	H	1	CH ₂ Ph
	55	4-CF ₃	н	н	H	1	CO ₂ Et
40	56	4-01	H	H	H	1	CO ₂ Et
40	57	4-8r	H	Н	H	1	CO ₂ Me
	58	4-8r	H	H	Н	1	CO ₂ Et
	59	4-C1	н	Н	H	1	CO ₂ He
45	60	4-c1	5-0Me	н	H	1	н
	61	4-CF ₃	Н	H	H	1	CO ₂ Ph
	62	4-CF ₃	Н	H	co ₂ -t-8u	1	н
50	63	4-SMe	5-C1	H	H	1	H
	64	4-C0_Et	5-C1	H	H	1	H

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TABLE 19 (continued)

	CHIPD	R ₁	<u>R</u> 2	<u>J</u>	¥	Ŧ	B
5	65	4-SMe	5-C1	н	н	1	CO,Ne
	66	4-C0 ₂ Et	5-C1	н	H	1	CO, He
		4-CF3	Н	н	н	2	н
10	68	4-CF ₃	Н	н	н	2	Me
	69	4-C1	н	н	н	2	н
	70	4-C1	н	H	н	2	Нe

TABLE 20

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R ₂	REBUTE NO SECULATION OF CONTRACT OF CONTRA
	_

	R ₁	R ₂		••	R ₈	R ₈
CHPD	<u></u>	_2	¥	Ā		_8
71	4-CF ₃	H	H	0	H	H
72	4-CF ₃	H	H	S	H	H
73	3,4-di-C1	H	Н	0	H	H
74	4-Br	H	H	0	H	H
75	3-CF ₃	H	H	0	H	H
76	4-C1	H	H	0	H	H
77	3-C1	H	H	0	H	H
78	3,4-di-C1	H	H	S	H	H
79	4-Br	H	H	S	H	H
80	3-CF ₃	H	H	S	H	H
81	4-C1	H	H	S	H	H
82	3-C1	H	H	\$	H	H
83	3-CF ₃	H	H	20(a)	H	H
84	4-CF3	H	H	so ₂	н	H
85	4-CF ₃	H	H	\$0(a)	H	H
86	3,4-di-C1	H	H	so ₂	H	H
87	4-Br	H	H	\$0(a)	H	H
88	4-Br	H	H	SO(B)	H	H
89	3,4-di-C1	H	H	\$0(a)	H	H
90	3,4-di-Cl	H	H	SO(B)	H	Н
91	4-C1	H	н	S O(a)	H	H
92	4-C1	H	н	SO(B)	H	H
93	3-01	н	н	\$0(a)	н	н
94	3-C1	H	H	SO(B)	H	H
95	4-Br	H	Н	so ₂	H	H
96	3-CF ₃	H	н	so ₂	н	н
97	4-C1	H	Н	so	H	H

TABLE 20 (continued)

5	CMPD	<u>R</u> 1	R ₂	ĭ	¥	R 8	8 g
	98	3-01	Н	H	80 ₂	H	H
10	99	3-CF ₃	н	H	80(8)	Н	H
10	100	4-CF3	5-C1	н	s	н	н
	101	3-CF3	5-C1	H	S	н	н
	102	4-C1	5-C1		S	H	н
15	103	3-C1	5-C1		S	H	H
	104	4-Br	5-C1		S	H	H
	105	3,4-di-Cl	5-01		S	H	H
	106	4-CF ₃	6-C1		S	H	H
20	107	3-CF ₃	6-C1	Н	S	Н	H
	108	4-C1	6-C1		S	H	H
	109	3-C1	6-C1		S	H	H
•	110	4-Br	6-C1		S	H	H
	111	3,4-di-Cl			S	H	H
25		4-CF ₃	5-01		80 ₂	Н	H
	113	3-CF ₃	5-C1		so ₂	H	H
	114	4-C1	5-C1		80 ₂	Н	H
30	115	3-C1	5-C1	Н	so ₂	Н	H
	116	4-Br	5-C1	H	so ₂	H	H
	117	3,4-di-Cl	5-C1	H	so ₂	H	H
	118	4-CF ₃	6-Cl	Н	so ₂	H	H
35	119	3-CF ₃	6-C1	H	80 ₂	H	H
	120	4-C1	6-C1	H	so ₂	H	H
	121	3-C1	6-C1	H	80 ₂	H	H
40	122	4-Br	6-C1	H	50 ₂	H	H
	123	3,4-di-Cl	6-C1	н	so ₂	H	H
	124	4-CF ₃	5-C1	н	80(a)	H	H
4 5	125	4-CF ₃	5-C1	H	SO(B)	H	H
	126	3-CF ₃	5-C1	н	(D)03	H	H
	127	3-CF ₃	5-C1	. н	50(8)	H	H
	128	4-C1	5-C1	H	50(a)	H	H
50	129	4-C1	5-C1		50(8)	H	H
	130	3-C1	5-C1		80(a)	H	H
	131	3-Cl	5-C1		80(8)	H	H
	132	4-Br	5-C1		50(a)	Н	H
	133	4-Br	5-C1	. н	50(8)	Н	H

83

TABLE 20 (continued)

	CMPD	$\frac{R_1}{}$	R ₂	¥	<u>v</u>	R ₈	R ₈
5	134	3,4-di-Cl	5-C1	н	SO(a)	н	H
	135	3.4-di-Cl	5-C1	H	SO(B)	H	H
	136	4-CF ₃	6-C1	H	50(a)	н	H
10	137	4-CF ₃	6-C1	Н	SO(B)	Н	H
	138	3-CF ₃	6-C1	H	SO(a)	H	H
	139	3-CF ₃	6-C1	H	SO(B)	Н	н
	140	4-C1	6-C1	H	SO(a)	H	H
15	141	4-Cl	6-C1	Н	SO(B)	H	H
	142	3-Cl	6-C1	H	SO(a)	H	H
	143	3-Cl	6-C1	H	SO(A)	H	H
	144	4-Br	6-C1	H	SO(a)	H	н
	145	4-Br	6-C1	H	SO(B)	H	H
20	146	3,4-di-Cl	6-C1	H	SO(a)	H	H
	147	3,4-di-Cl	6-C1	H	50(B)	H H	H H
	148	4-CF ₃	5-Br	H	S		
	149	3-CF ₃	5-Br	H	S	Н	H
25	150	4-Cl	5-Br	H .	8	H	H
	151	3-C1	5-Br	H	5	H	н
	152	4-Br	5-Br	H	S	H	H
	153	3,4-di-Cl	5-Br	H	5	H	H
	154	4-CF ₃	н	H	SO(B)	H	H
30	155	4-CF ₃	H	H	5	H	H
	156	4-CF ₃	н	H	S	H	H
	157	4-CF ₃	H	H	5	H	H
35	158	4-CF ₃	5-C1	C(O)Me	S	H	H
	159	4-CF3	5-Cl	CO ₂ Me	S	H	H
	160	4-CF ₃	5-Cl	Me	S	H	H
	161	4-C1	H	C(O)Me	S	H	H
40	162	4-C1	H	Me	S	H	H
	163	4-Cl	5-C1	C(O)Me	S	H	H
	164	4-C1	5-C1	Me	S	H	H
	165	4-CF ₃	Н	H	S	Me	Me
45	166	4-Cl	H	H	S	Me	Me
	167	4-Br	H	H	S	Me	Me
	168	4-Cl	5-Cl	H	5	Me	Me
	169	4-Br	5-Cl	H	5	Me	Мe
	170	4-CF ₃	H	Н	S	Me	Н
50	171	4-CF ₃	Н	Н	S	H	Me

TABLE 20 (continued)

_	CMPD	<u>R</u> 1	<u>R</u> 2	¥	<u>v</u>	R ₈	3 8
5	172	4-C1	н	н	S	He	н
	173	4-G1	H	H	Š	н	Xe
	174	4-Br	н	Н	S	He	H
	175	4-Br	H	н	S	H	He
10	176	4-CF ₃	5-C1	H	S	He	н
	177	4-CF3	5-C1	н	s	H	He
	178	4-C1	5-C1	н	S	Жe	н
	179	4-C1	5-C1	Н	S	H	Me
15	180	4-Br	5-C1	H	S	Нe	H
-	181	4-Br	5-C1	н	S	н	Нe

Example 9

Fall Armyworm

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Test units, each consisting of an 8-ounce plastic cup containing a layer of wheat germ diet, approximately 0.5 cm thick, were prepared. Ten third-instar larvae of fall armyworm (Spodoptera frugiperda) were placed into each cup. Solutions of each of the below-listed test compounds (acetone distilled water 75:25 solvent) were sprayed onto the cups, a single solution per cup. Spraying was accomplished by passing the cups, on a conveyor belt, directly beneath a flat fan hydraulic nozzle which discharged the spray at a rate of 0.5 pounds of active ingredient per acre (about 0.55 kg/ha) at 30 p.s.i. The cups were then covered and held at 27°C and 50% relative humidity for 72 hours, after which time mortality readings were taken. Of the compounds tested on fall armyworm, the following resulted in greater than or equal to 80% mortality: 3, 4, 5, 7, 15, 17, 20, 22, 23, 25, 45, 59, 72, 100, 125, 148, 158, 159, 160, 165, 166, 168, 169, 170, 173, 177, 179, 181.

Example 10

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European Corn Borer

Test units, each consisting of an 8-ounce plastic cup containing a one-inch square of wheat germ/soyflour diet were prepared. Five third-instar larvae of the European corn borer (Ostrinia nubilalis) were placed into each cup. The test units were sprayed as described in Example 9 with individual solutions of the below-listed compounds. The cups were then covered and held at 27°C and 50% relative humidity for 48 hours, after which time mortality readings were taken. Of the compounds tested on European corn borer, the following resulted in greater than or equal to 80% mortality: 2, 3, 4, 5, 15, 22, 23, 29, 45, 47, 53, 54, 55, 56, 58, 59, 71, 72, 81, 84, 91, 95, 97, 100, 165, 177.

Example 11

Southern Corn Rootworm

Test units, each consisting of an 8-ounce plastic cup containing 1 sprouted corn seed, were prepared. Set of three test units were sprayed as described in Example 9 with individual solutions of the test compounds. After the spray on the cups had dried, five third-instar larvae of the southern corn rootworm (Diabrotica undecimpunctata howardi) were placed into each cup. A moistened dental wick was inserted into each cup to prevent drying and the cups were then covered. The cups were then held at 27°C and 50% relative humidity for 48 hours, after which time mortality readings were taken. Of the compounds tested on Southern rootworm, the following resulted in greater than or equal to 80% mortality: 1, 2, 3, 4, 5, 7, 15, 16, 17, 20, 21, 22, 23, 25, 27, 29, 35, 36, 45, 47, 53, 54, 56, 58, 59, 72, 76, 79, 81, 84, 91, 95, 97, 100, 124, 125, 132, 133, 148, 154, 157, 158, 160, 162, 165, 166, 168, 169, 170, 172, 173, 174, 175, 180.

15 Example 12

Boll Weevil

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Five adult boll weevils (<u>Anthonomus grandis</u>) were placed into each of a series of 9-ounce cups. The test procedure employed was then otherwise the same as in Example 9. Mortality readings were taken 48 hours after treatment. Of the compounds tested on boll weevil, the following resulted in greater than or equal to 80% mortality: 1, 2, 3, 4, 5, 7, 14, 15, 22, 23, 35, 45, 47, 53, 58, 59, 61, 67, 71, 72, 76, 81, 84, 91, 97, 104, 165, 177.

Example 13

Aster Leafhopper

Test units were prepared from a series of 12-ounce cups, each containing oat (Avena sativa) seedlings in a 1-inch layer of sterilized soil. The test units were sprayed as described in Example 9 with individual solutions of the below-listed compounds. After the oats had dried from the spraying, between 10 and 15 adult aster leafhoppers (Mascrosteles fascifrons) were aspirated into each of the covered cups. The cups were held at 27°C and 50% relative humidity for 48 hours, after which time mortality readings were taken. Of the compounds tested on aster leafhopper, the following resulted in greater than or equal to 80% mortality: 2, 3, 4, 5, 7, 15, 16, 20, 22, 23, 25, 45, 47, 53, 59, 72, 79, 100, 104, 157, 158, 159, 160, 162, 165, 166, 168, 169, 171, 172, 173, 174, 175, 177, 179, 181.

Example 14

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Tobacco Budworm

The test procedure of Example 9 was repeated on third-instar larvae of the tobacco budworm (<u>Heliothis virescens</u>) except that mortality was assessed at 48 hours. Of the compounds tested on tobacco budworm, the following resulted in greater than or equal to 80% mortality: 5, 7, 15, 23, 25, 39, 100, 160, 165, 168, 177.

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Example 15

Soil Control of Southern Corn Rootworm

Test units, each consisting of a three inch cylindrical pot containing 200 grams of moist sassafras foam soil and 2 kernels of Dekalb T1100 corn were prepared. Spraying was accomplished by passing the pots, on a conveyer belt, directly beneath a flat fan hydraulic nozzle which discharged the spray at a rate of 0.25 pounds of active ingredient per acre (about 2 ppm soil concentration) at 30 p.s.i. After the test units had been sprayed, the corn seed was covered with soil and ten third-instar southern corn rootworm larvae (Diabrotica undecimpunctata howardi) were added to each unit. Test units were held at 24 degree C, 70% relative humidity and under a photo cycle of 16 hours of light/8 hours of darkness for 7 days, after which time control readings were taken. Control was measured by root injury ratings of 1 to 5, where 1 was no injury and 5 was complete destruction of the root system. Percent of corn seedlings that emerged also was utilized to measure control. The results are tabulated below.

15	Compound	Root Rating	Emergence
	2	1.5	100
	3	1.0	88
	5	1.0	100
20	15	1.3	88
	20	1.6	75
	22	2.0	75
	23	1.0	75
	29	1.0	100
25	59	1.1	100

Claims

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1. A compound of the formula

wherein Q is

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A is a 1, 2 or 3-atom bridge having 0 to 3 carbon atoms and 0 to 1 oxygen atom, NR₆ group, or S(O)_q group, wherein each carbon individually can be substituted with 1 to 2 substituents selected from 1 to 2 halogen, C₂ to C₃ alkyl, C₂ to C₄ alkoxycarbonyl or phenyl optionally substituted with 1 to 3 substituents selected from W and one of the carbon atoms can be C(O) or C(S);

B is H, C₁ to C₅ alkyl, C₁ to C₆ haloalkyl, C₂ to C₅ alkenyl, C₂ to C₅ haloalkenyl, C₂ to C₆ alkynyl, OR₇, C₂ to C₅ alkoxyalkyl, C₂ to C₅ cyanoalkyl, C₃ to C₈ alkoxycarbonylalkyl, CO₂R₃, C(O)R₃, C(O)NR₃R₄, C(S)-

 NR_3R_4 , $C(S)R_3$, $C(S)SR_3$, phenyl, phenyl substituted by $(R_5)_p$, benzyl, or benzyl substituted with 1 to 3 substituents independently selected from W:

J is H, C- to C4 alkyl or phenyl optionally substituted with W;

K is H or CH3;

W is halogen, CN, NO₂, C₁ to C₂ alkyl. C₁ to C₂ haloalkyl, C₁ to C₂ alkoxy. C₂ to C₃ haloalkoxy. C₄ to C₂ alkylsulfonyl or C₄ to C₃ haloalkylsulfonyl;

R. R₂ and R₅ are independently R₃, halogen, CN, N₃, SCN, NO₂, OR₃, SR₃, SO₂R₃, OC(O)R₂, OSO₂R₃, CO₂R₃, C(O)R₃R₄, SO₂NR₃R₄, NR₃R₄, NR₄C(O)R₃, OC(O)NHR₂, NR₄C(O)NHR₃, NR₄C(O)R₃, or when m, n or p is 2, R₁, R₂ or R₅ can be taken together to form a 5 or 6 membered fused ring as -OCH₂O-, -OCH₂CH₂O-, or -CH₂CH₂O-, each of which can be substituted with 1 to 4 halogen atoms or 1 to 2 methyl groups;

R₃ is H, C₁ to C₆ alkyl, C₁ to C₅ haloalkyl, C₂ to C₅ alkenyl, C₂ to C₅ haloalkenyl, C₂ to C₅ alkynyl, C₂ to C₅ haloalkynyl, C₂ to C₅ alkoxyalkyl, C₂ to C₆ alkylthioalkyl, C₁ to C₆ nitroalkyl, C₂ to C₅ cyanoalkyl, C₃ to C₆ alkoxycarbonylalkyl, C₃ to C₆ halocycloalkyl, phenyl, benzyl, or phenyl or benzyl substituted with 1 to 3 substituents independently selected from W;

 R_4 is H, C_1 to C_4 alkyl or R_3 and R_4 can be taken together as $(CH_2)_4$, $(CH_2)_5$ or $(CH_2CH_2OCH_2CH_2)$; R_6 is H, C_1 to C_4 alkyl, C_1 to C_4 haloalkyl, C_2 to C_4 alkenyl, C_2 to C_4 haloalkenyl, phenyl optionally substituted with W;

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R7 is H, C1 to C4 alkyl, C2 to C4 alkenyl, C2 to C4 alkynyl, C2 to C4 alkylcarbonyl, C2 to C4 alkoxycarbonyl, C2 to C4 alkylsulfonyl;

X is O or S;

n is 1 to 4;

m is 1 to 5:

p is 1 to 3;

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35

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25 q is 0 to 2; and

Y is H, C+ to C₅ alkyl, benzyl, C₂ to C₅ alkoxycarbonyl, CHO, C₂ to C₅ alkylcarbonyl, C₂ to C₅ alkoxycarbonyl, C+ to C₅ haloalkylcarbonyl, C+ to C₅ haloalkylthio, phenylthio, or phenylthio substituted with 1 to 3 substituents independently selected from W.

2. A compound according to Claim 1 wherein Q is

$$(R_{2})_{n} \xrightarrow{(R_{8})_{g}} (R_{2})_{n} \xrightarrow{(R_{8})_{g}} (R_{8})_{g}$$

$$(R_{2})_{n} \xrightarrow{(R_{2})_{n}} K$$

$$Q-1$$

$$Q-2$$

(R₂)
$$f$$

(R₂) f

(R₂

10

5

X is O or S;

X' is O or S:

t is 0, 1 or 2;

V is O, S(O)q, or NR6;

Z is O or NR6.

 R_8 is H, halogen, C_1 - C_6 alkyl, C_2 - C_4 alkoxycarbonyl, phenyl or phenyl substituted by 1-3 substituents selected from W; and

g is 0, 1 or 2.

3. A compound according to Claim 2 wherein

B is H, C₂ to C₅ alkyl, C₁ to C₅ haloalkyl, C₂ to C₅ alkenyl, C₂ to C₆ alkoxyalkyl, C₃ to C₈ alkoxyalkyl, C₀ to C₅ alkoxyalkyl, C₂ to C₅ alkoxyalkyl, C₂ to C₅ alkoxyalkyl, C₃ to C₈ alkoxyalkyl, C₂ to C₅ alkoxyalkyl, C₃ to C₈ alkoxyalkyl, C₂ to C₅ alkoxyalkyl, C₃ to C₈ alkoxyal

J is H or CH3;

R₃ is H, C₁ to C₄ alkyl, C₁ to C₂ haloalkyl, C₃ to C₄ alkenyl, C₃ to C₄ haloalkenyl, propargyl, phenyl, benzyl, or phenyl or benzyl substituted with one of F. Cl, Br, CF₃ or NO₂;

R₆ is H. C· to C₄ alkyl. allyl or propargyl:

n is 1 or 2:

p is 1 or 2:

m is 1 to 3; and

Y is H. C- to C4 alkyl. SCH3, SCCl3, SO₂CH3, SC₆H5, 2-NO₂-C₆H4S, C(O)CH3, C(O)CF3; CO₂CH3 or CO₂CH₂CH₃.

4. A compound according to Claim 3 wherein

B is H. C. to C_4 alkyl. C_3 - C_4 alkoxycarbonylalkyl. CO_2R_3 , $C(O)R_3$, phenyl or phenyl substituted by $(R_5)_p$; J is H;

35 Kis H:

R₁ is H, halogen, CN, SCN, NO₂, OR₃, SR₃, SO₂R₃, CO₂R₃, C(O)R₃ or R₃ with one R₁ substituent in the 4-position, or when m is 2 then R₂ can be taken together as -CH₂C(CH₃)₂O₂-, OCH₂CH₂O₃-, OCF₂CF₂O-or -CF₂CF₂O-to form a 5 or 6 membered fused ring;

R₂ and R₅ are H, R₃, halogen, CN, SCN, NO₂, OR₃, SR₃, SO₂R₃, OC(O)R₃, OSO₂R₃, CO₂R₃, C(O)R₃, C-(O)NR₃R₄, SO₂NR₃R₄ or NR₃R₄;

R₃ is C·-C₄ alkyi, C·-C₂ haloalkyl, C₃-C₄ alkenyl or propargyl;

R4 and R8 are H or CH3;

X is O;

X' is O;

m is 1 or 2:

t is 1; and

q is 0.

5. A compound according to Claim 4 wherein

R₁ is Cl, F, Br, CF₃, CN, OCF₃, OCF₂H or SCF₂H;

 H_2 is H, Cl, F, Br, CN, CF₃, CH₃, OCF₂H, OCF₃, SCH₃, SCF₂H, SO₂CH₃ or NO₂; H_6 is H or CH₃;

B is H, CH₃, CH₂CH₃, CO₂CH₃, CO₂CH₂CH₃, phenyl or phenyl substituted by halogen, CN, CF₃ or NO₂, and;

Y is H.

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- 6. A compound according to Claim 5 wherein Q is Q-1.
- 7. A compound according to Claim 5 wherein Q is Q-2.
- 8. A compound according to Claim 6:

methyl 3,3a,4.5-tetrahydro-2-[[4-(trifluoromethyl)phenylamino]carbonyl]-2H-benz[g[indazole-3a-carboxylate.

- 9. A compound according to Claim 6: methyl 7-chloro-3,3a,4,5-tetrahydro-2-[[4-trifluoromethyl]phenyiaminc]carbonyi]-2H-benz[g]indazole-3a-carboxylate.
 - 10. A compound according to Claim 6:
- 7-chloro-3,3a.4.5-tetrahydro-N-[4-(trifluoromethyl)phenyl]-2H-benz[g]indazoie-2-carboxamide.
 - 11. A compound according to Claim 6: methyl 6-fluoro-3,3a.4,5-tetrahydro-2-[[4-(trifluoromethyl)phenylamino]carbonyl]-2H-benz[g]indazole-3a-carboxylate.
- 12. A composition comprising an insecticidally effective amount of a compound according to any one of Claims 1 to 11 and an agriculturally suitable carrier therefor.
 - 13. A method for controlling insects comprising contacting them with an insecticidially effective amount of a compound according to any one of Claims 1 to 11.

20 25 30 35 40 **4**5 50 55